

Name \_\_\_\_\_

**SHORT ANSWER.** Write the word or phrase that best completes each statement or answers the question.

- 1) What is the solvent in an aqueous solution?

**ESSAY.** Write your answer in the space provided or on a separate sheet of paper.

- 2) Explain why water is a good solvent for ionic substances.

**MULTIPLE CHOICE.** Choose the one alternative that best completes the statement or answers the question.

- 3) Of the species below, only \_\_\_\_\_ is
- not
- an electrolyte.

A) HCl                      B) Ar                      C) NaCl                      D) KOH                      E) Rb<sub>2</sub>SO<sub>4</sub>

- 4) What is the molarity of an aqueous solution containing 75.3 g of glucose (C
- <sub>6</sub>
- H
- <sub>12</sub>
- O
- <sub>6</sub>
- ) in 35.5 mL of solution?

A) 0.197                      B) 2.12                      C) 3.52                      D) 1.85                      E) 11.8

- 5) The concentration of species in 500 mL of a 2.104 M solution of sodium sulfate is \_\_\_\_\_ M sodium ion and \_\_\_\_\_ M sulfate ion.

A) 4.208, 2.104                      B) 1.052, 1.052                      C) 2.104, 1.052                      D) 2.104, 4.208                      E) 2.104, 2.104

- 6) What is the molarity of a solution prepared by diluting 43.72 mL of 1.005 M aqueous K
- <sub>2</sub>
- Cr
- <sub>2</sub>
- O
- <sub>7</sub>
- to 500 mL?

A) 0.879                      B) 87.9                      C) 0.0879                      D) 0.0115                      E) 0.0218

- 7) When aqueous solutions of \_\_\_\_\_ are mixed, a precipitate forms.

A) NaI and KBr  
B) Li<sub>2</sub>CO<sub>3</sub> and CsI  
C) K<sub>2</sub>SO<sub>4</sub> and CrCl<sub>3</sub>  
D) NiBr<sub>2</sub> and AgNO<sub>3</sub>  
E) KOH and Ba(NO<sub>3</sub>)<sub>2</sub>

- 8) What are the spectator ions in the reaction between KOH (aq) and HNO
- <sub>3</sub>
- (aq)?

A) H<sup>+</sup> and NO<sub>3</sub><sup>-</sup>  
B) OH<sup>-</sup> only  
C) K<sup>+</sup> and NO<sub>3</sub><sup>-</sup>  
D) K<sup>+</sup> and H<sup>+</sup>  
E) H<sup>+</sup> and OH<sup>-</sup>**SHORT ANSWER.** Write the word or phrase that best completes each statement or answers the question.

- 9) What is the concentration of HCl in a 0.100 M solution of HCl? Why?

# Ap Chemistry Chapter 4 Test

**John T. Moore, Richard H. Langley**



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