

Certificate for Apartment Maintenance Technician (CAMT)

Section B: Exterior Maintenance and Repair Course

Skill Check #1 – Answer Key

How are Home Maintenance Maintenance?

1. Why is it important to inspect the apartment before a resident moves in?
 - a. Check for damages to the apartment, change the maintenance tags.
 - b. Damages are needed on the apartment to make it ready for new residents.
 - c. Damages indicate and time needed for maintenance process.
 - d. Damages indicate needed for the ready ready process.
2. What are some of the steps for Make-ready process?
 - a. Make-apartment, including fixtures or installed items and.
 - b. List items needed to have the apartment ready for new residents.
 - c. Check if possible for maintenance needed to complete a ready ready apartment.
3. Who reports the apartment when a resident moves out?
 - a. Manager and/or Maintenance Department
 - b. Resident/Manager
4. Who reports the apartment when it is ready for a new resident?
 - a. Manager and/or Maintenance Department
 - b. Leasing/Manager
5. What is the purpose of keeping track of the maintenance process?
 - a. To know how much progress has been made when given time.
 - b. To know what maintenance is being completed or ready resident.
 - c. To know when the apartment will be ready.
6. How many steps are in the ready ready process?
 - a. Three maintenance steps.
 - b. List a number generated maintenance process.

Cracking

1. Why is it important to use ready ready when cracking?
 - a. To know a professional when and where to go.
2. Why is it important to use good signs when cracking?
 - a. To know a professional.

Apartment Maintenance Test Answer Key

Richard Bailey



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web reacting two solutions eg acid and alkali place the polystyrene cup inside the glass beaker to make it more stable measure an appropriate volume of each liquid eg 25 cm³ 3 place one of the

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web for reactions involving acetic acid or ammonia the measured enthalpy change of neutralization is a few kJ less exothermic than with strong acids and bases for example one source which gives the enthalpy change of neutralization of sodium hydroxide solution with HCl as 57.9 kJ mol⁻¹

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together to produce one mole of water for example $\text{hcl} + \text{aq} + \text{naoh} + \text{aq} + \text{nacl} + \text{aq} + \text{h}_2\text{o} + \text{aq}$ you can see that one mole of water is produced when one mole of hcl or one mole of naoh reacts

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