

Microorganisms WebQuest

protists, fungi, bacteria and
viruses

A tablet displaying a webquest worksheet. The worksheet has fields for Name, Date, and Score. The title is 'MICROORGANISMS webquest'. Below the title is an instruction: 'Use the videos below to find the answers to the following questions. You should watch the entire video. Know that the questions are written in the order that they will be discussed through the video.' There are two video thumbnails: 'Protists and Fungi' and 'Bacteria and Viruses'. Below the thumbnails is a table with five rows of questions and a column for the answer.

Name: _____ Date: _____ Score: _____

MICROORGANISMS webquest

Use the videos below to find the answers to the following questions. You should watch the entire video. Know that the questions are written in the order that they will be discussed through the video.

[Protists & Fungi](#)
[https://www.youtube.com/watch?v=K7C4m9m3z0](#)

[Bacteria and Viruses](#)

1. Are protists prokaryotes or eukaryotes?	1 pt
2. What are 3 things protists might have in order to survive?	3 pt
3. If you wanted to find a protist, where would you look?	1 pt
4. Are protists multicellular or unicellular?	
5. Protists do a lot of good in the world. Name some of these things.	1 pt

Microorganisms Webquest

Lei Shi



Microorganisms Webquest:

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web 1 define each remember oil rig oxidation is loss of e reduction is gain of e a oxidation b reduction c oxidizing agent d reducing agent 2 write half reactions for each of the following atoms or ions label each as oxidation or reduction use the activity series chart a al b ba2 c br 2 d ca e ga3 f h 2 g h

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oxidation $\text{Co} \rightarrow \text{Co}^{2+} + 2\text{e}^{-}$ reduction $\text{Fe}^{2+} + 2\text{e}^{-} \rightarrow \text{Fe}$ 24 3 $\text{Ag} + \text{Ni}^{3+} \rightarrow \text{Ni} + 3\text{Ag}$ oxidation reduction 25 $\text{Cu} + \text{Pb}^{2+} \rightarrow \text{Pb} + \text{Cu}$ oxidation reduction 26 $\text{O}_2 + 2\text{H}_2 \rightarrow 2\text{H}_2\text{O}$

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 $5\text{CO} + \text{I}_2 \rightarrow 5\text{CO}_2 + 2\text{I}$ $5\text{CO} + \text{I}_2 \rightarrow 5\text{CO}_2 + 2\text{I}$

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