

GET TO KNOW YOUR PERIODIC TABLE

1 (1A)	2 (2A)												13 (3A)	14 (4A)	15 (5A)	16 (6A)	17 (7A)	18 (8A)
1 H 1.00794													B 10.811	C 12.011	N 14.007	O 15.999	F 18.998	Ne 20.180
2 Li 6.941	Be 9.012											Al 26.982	Si 28.086	P 30.974	S 32.06	Cl 35.453	Ar 39.948	
3 Na 22.990	Mg 24.305	3 (3B)	4 (4B)	5 (5B)	6 (6B)	7 (7B)	8 (8B)	9 (9B)	10 (10B)	11 (11B)	12 (12B)	Ga 69.723	Ge 72.63	As 74.922	Se 78.96	Br 79.904	Kr 83.8	
4 K 39.102	Ca 40.08	Sc 44.956	Ti 47.88	V 50.942	Cr 51.996	Mn 54.938	Fe 55.847	Co 58.933	Ni 58.71	Cu 63.54	Zn 65.37	Ga 69.72	Ge 72.63	As 74.922	Se 78.96	Br 79.904	Kr 83.8	
5 Rb 85.47	Sr 87.62	Y 88.906	Zr 91.224	Nb 92.906	Mo 95.94	Tc 98	Ru 101.07	Rh 102.905	Pd 106.4	Ag 107.87	Cd 112.4	In 114.82	Sn 118.71	Sb 121.76	Te 127.6	I 126.905	Xe 131.3	
6 Cs 132.905	Ba 137.34	La 138.905	Hf 178.49	Ta 180.948	W 183.85	Re 186.2	Os 190.2	Ir 192.22	Pt 195.08	Au 196.967	Hg 200.59	Tl 204.37	Pb 207.19	Bi 208.98	Po 209	At 210	Rn 222	
7 Fr 223	Ra 226	Ac 227	Rf 261	Db 262	Sg 263	Bh 264	Hs 265	Mt 266	Dun 267	Uuu 272	Uub 277		Uuq 289		Uuh 288		Uuo 289	

Ce 140.12	Pr 140.907	Nd 144.24	Pm 145	Sm 150.36	Eu 151.96	Gd 157.25	Tb 158.925	Dy 162.5	Ho 164.93	Er 167.26	Tm 168.934	Yb 173.04	Lu 174.967
Th 232.038	Pa 231.04	U 238.03	Np 237	Pu 244	Am 243	Cm 247	Bk 247	Cf 251	Es 252	Fm 257	Md 258	No 259	Lr 260

Periods and Groups

The horizontal rows in the table are referred to as **periods**, which are numbered from 1 to 7. The columns are referred to as **groups** or **families**. The numbering system for groups varies between textbooks and instructors. The newest numbering system keeps things simple; groups are numbered from 1 to 18 (as shown on your periodic table).

The “old school” numbering system separates the groups into two categories: “A” group elements (also known as main group elements) and “B” group elements (also known as transition elements or transition metals). The main groups make up the left and right sides of the periodic table and are labeled in order from 1A-8A. The transition metal groups are located in the middle of the periodic table and are labeled in a bizarre fashion that goes in this order: 3B, 4B, 5B, 6B, 7B, 8B, 1B, 2B. (To make matters even more confusing, “group” 8B is actually made up of *three* groups of elements!)

Needless to say, we’re going to focus on the newest numbering system, with groups numbered from 1-18. But, since some textbooks still use the old numbering system, it is important to be familiar with both. On your periodic table, please label the **main groups** with the old numbering system in addition to the new numbering system. (Group 1 = Group 1A, Group 2 = Group 2A, Group 13 = Group 3A, Group 14 = Group 4A, etc.) Don’t worry about re-labeling the transition metals. Use the diagram above as a guide.

Getting To Know The Elements Answer Key

A Loxley



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