

## Chemical Formula Writing Worksheet Solutions

Write chemical formulas for the compounds in each box. The names are found by finding the intersection between the cations and anions. Example: The first box is the intersection between the "zinc" cation and the "chloride" anion, so you should write "ZnCl<sub>2</sub>", as shown.

	<i>zinc</i>	<i>iron (II)</i>	<i>iron (III)</i>	<i>gallium</i>	<i>silver</i>	<i>lead (IV)</i>
<i>chloride</i>	<b>ZnCl<sub>2</sub></b>	<b>FeCl<sub>2</sub></b>	<b>FeCl<sub>3</sub></b>	<b>GaCl<sub>3</sub></b>	<b>AgCl</b>	<b>PbCl<sub>4</sub></b>
<i>acetate</i>	<b>Zn(C<sub>2</sub>H<sub>3</sub>O<sub>2</sub>)<sub>2</sub></b>	<b>Fe(C<sub>2</sub>H<sub>3</sub>O<sub>2</sub>)<sub>2</sub></b>	<b>Fe(C<sub>2</sub>H<sub>3</sub>O<sub>2</sub>)<sub>3</sub></b>	<b>Ga(C<sub>2</sub>H<sub>3</sub>O<sub>2</sub>)<sub>3</sub></b>	<b>Ag<sub>2</sub>C<sub>2</sub>H<sub>3</sub>O<sub>2</sub></b>	<b>Pb(C<sub>2</sub>H<sub>3</sub>O<sub>2</sub>)<sub>4</sub></b>
<i>nitrate</i>	<b>Zn(NO<sub>3</sub>)<sub>2</sub></b>	<b>Fe(NO<sub>3</sub>)<sub>2</sub></b>	<b>Fe(NO<sub>3</sub>)<sub>3</sub></b>	<b>Ga(NO<sub>3</sub>)<sub>3</sub></b>	<b>AgNO<sub>3</sub></b>	<b>Pb(NO<sub>3</sub>)<sub>4</sub></b>
<i>oxide</i>	<b>ZnO</b>	<b>FeO</b>	<b>Fe<sub>2</sub>O<sub>3</sub></b>	<b>Ga<sub>2</sub>O<sub>3</sub></b>	<b>Ag<sub>2</sub>O</b>	<b>PbO<sub>2</sub></b>
<i>nitride</i>	<b>Zn<sub>3</sub>N<sub>2</sub></b>	<b>Fe<sub>3</sub>N<sub>2</sub></b>	<b>FeN</b>	<b>GaN</b>	<b>Ag<sub>3</sub>N</b>	<b>Pb<sub>3</sub>N<sub>4</sub></b>
<i>sulfate</i>	<b>ZnSO<sub>4</sub></b>	<b>FeSO<sub>4</sub></b>	<b>Fe<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub></b>	<b>Ga<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub></b>	<b>Ag<sub>2</sub>SO<sub>4</sub></b>	<b>Pb(SO<sub>4</sub>)<sub>2</sub></b>

Write the formulas for the following compounds:

- 1) copper (II) chloride **CuCl<sub>2</sub>**
- 2) lithium acetate **LiC<sub>2</sub>H<sub>3</sub>O<sub>2</sub>**
- 3) vanadium (III) selenide **V<sub>2</sub>Se<sub>3</sub>**
- 4) manganese (IV) nitride **Mn<sub>3</sub>N<sub>4</sub>**
- 5) beryllium oxide **BeO**
- 6) sodium sulfate **Na<sub>2</sub>SO<sub>4</sub>**
- 7) aluminum arsenide **AlAs**
- 8) potassium permanganate **KMnO<sub>4</sub>**
- 9) chromium (VI) cyanide **Cr(CN)<sub>6</sub>**
- 10) tin (II) sulfite **SnSO<sub>3</sub>**
- 11) vanadium (V) fluoride **VF<sub>5</sub>**
- 12) ammonium nitrate **NH<sub>4</sub>NO<sub>3</sub>**

# Ionic Formula Writing Kit Answers

**Michael J. Padilla**



## **Ionic Formula Writing Kit Answers:**

**An Introduction to Chemistry** Michael Mosher, Paul Kelter, 2023-03-18 This textbook is written to thoroughly cover the topic of introductory chemistry in detail with specific references to examples of topics in common or everyday life It provides a major overview of topics typically found in first year chemistry courses in the USA The textbook is written in a conversational question based format with a well defined problem solving strategy and presented in a way to encourage readers to think like a chemist and to think outside of the box Numerous examples are presented in every chapter to aid students and provide helpful self learning tools The topics are arranged throughout the textbook in a traditional approach to the subject with the primary audience being undergraduate students and advanced high school students of chemistry

Questions set at the examinations College Entrance Examination Board, 1923 Questions Set at the Examinations, 1922 Questions Set at the Comprehensive Examinations of ..., 1924 *Chemistry at a Glance* Roger Owen, Sue King, 2005-06-27 This book aims to cover the specifications of the main examination boards for GCSE Double Science GCSE Single Science and the core content of GCSE Chemistry Where relevant Key Stage 3 material is summarized as an introduction to GCSE topics This serves as revision of work done prior to Key Stage 4 and a foundation for GCSE studies The book is **Chemistry** James E. Brady, Fred Senese, 2004-02-04 Publisher Description **Chemistry** Neil D. Jespersen, Alison Hyslop, 2021-11-02 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry Simultaneously examples and questions have been updated to align them with career concepts relevant to the environmental engineering biological pharmaceutical and medical sciences Providing students with transferable skills with a focus on integrating metacognition and three dimensional learning into the text When students know what they know they are better able to learn and incorporate the material Providing a total solution through New WileyPLUS by fully integrating the enhanced etext with online assessment answer specific responses and additional practice resources The 8th edition continues to emphasize the importance of applying concepts to problem solving to achieve high level learning and increase retention of chemistry knowledge Problems are arranged in an intuitive confidence building order Prentice Hall Chemistry, 2000 **Chemistry, an Introduction** Stanley M. Cherim, Leo E. Kallan, 1980 Basic Chemistry Steven S. Zumdahl, 2004 Description Not Yet Available **Prentice Hall Science Explorer** Michael J. Padilla, 2002 **Numerical Chemistry for Competitions** Anu Sharma, 2009 An ideal book for the students of XI and XII CBSE ISC and the State Boards who are using Core Curriculum and also useful for the students preparing for various Engineering Medical Entrance Examinations **Science** John Michels (Journalist), 1903 Vols for 1911 13 contain the Proceedings of the Helminthological Society of Washington ISSN 0018 0120 1st 15th meeting *Lutheran*

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*Mechanic and Mirror of Science and Art* ,1885

## **Ionic Formula Writing Kit Answers** Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Ionic Formula Writing Kit Answers**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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