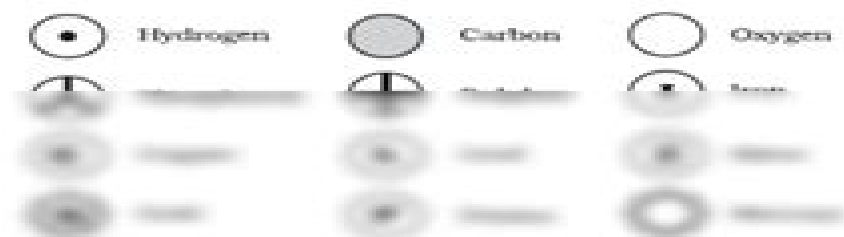




FUNDAMENTAL CHEMICAL LAWS & DALTON'S ATOMIC THEORY

The *Modern Atomic Theory*, proposed about 1803 by the English chemist John Dalton is a fundamental concept that states that all elements are composed of atoms. This theory is based on the following sets of postulates:

- All matter consists of indivisible particles called atoms.
- All the atoms of an element are identical in mass and size, and are different from the atoms of another element. Dalton used the different shapes or figures to represent different elements, as follows:



- Compounds are composed of atoms from two or more elements combined in definite ratio with whole number ratio.

Laws Of Chemical Combination Daltons Atomic Theory

David Baud



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