

CONFORMATIONAL ANALYSIS: PRACTICE EXERCISES

10. Draw a Newman projection of the most stable conformation of 2-methylpropane.

11. The structures below are



- A) staggered.  
 B) nonstaggered.  
 C) eclipsed.  
 D) structural isomers.  
 E) both C and D.

12. Define the term *conformation*.

13. Draw a *staircase* molecule along the  $C_2-C_3$  bond and provide a *Newman* projection of the lowest energy conformation.

14. Provide a representation of the *gauche* conformation of butane.

15. Among the butane conformers, which occur at energy minima on a graph of potential energy versus dihedral angle?

- A) gauche only.  
 B) eclipsed and totally eclipsed.  
 C) anti and anti.  
 D) eclipsed only.  
 E) anti only.

16. Which of the following substituents exhibits the greatest negative heat of combustion per  $\text{CH}_3$  group?

- A) cyclohexane     B) cyclopentane     C) cyclohexene     D) cyclobutane     E) cyclopropane

17. Which of the following correctly ranks the substituents in order of increasing the heat per methylene ( $\text{CH}_2$ ) group?

- A) cyclohexane < cyclopentane < cyclohexene < cyclopropane  
 B) cyclohexane < cyclopentane < cyclohexene < cyclopropane  
 C) cyclohexane < cyclohexene < cyclopentane < cyclopropane  
 D) cyclohexane < cyclopropane < cyclopentane < cyclohexene  
 E) cyclopropane < cyclohexane < cyclopentane < cyclohexene

18. Describe the number of angles which will be formed when propan-2-ol is cyclopropaned.

# Conformational Analysis Practice Exercises

**Ying-Ying Zheng**



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