

## Past Simple / Past Continuous (2)

1.- Complete the sentences with the verbs in brackets into the correct tense Past Simple or Past Continuous.



- 1.- It \_\_\_\_\_ (snow) when I \_\_\_\_\_ (leave) home this morning.
- 2.- It was a sunny afternoon and people \_\_\_\_\_ (sit) on the grass in the park. Then suddenly it \_\_\_\_\_ (start) to rain.
- 3.- A: I tried to explain my problem to her.  
B: \_\_\_\_\_ (she/listen)?
- 4.- My brother \_\_\_\_\_ (talk) on the phone when I arrived, but when he \_\_\_\_\_ (see) me, he \_\_\_\_\_ (finish) the call. Perhaps he thought that I \_\_\_\_\_ (listen) to his conversation.
- 5.- I nearly had an accident today. A car \_\_\_\_\_ (come) towards me, but I moved quickly out of the way and fortunately nothing \_\_\_\_\_ (happen).
- 6.- A: Which hotel \_\_\_\_\_ (you/stay) in when you lost your passport?  
B: I don't remember. I \_\_\_\_\_ (visit) many places during my European tour and I \_\_\_\_\_ (stay) in many different hotels.
- 7.- I \_\_\_\_\_ (move) to Hastings in 1999. I \_\_\_\_\_ (live) there when I met them.
- 8.- \_\_\_\_\_ (many people / wait) for the 9:15 bus last night?
- 9.- I \_\_\_\_\_ (dance) with my girlfriend when the disc jockey played our favourite song.
- 10.- I \_\_\_\_\_ (study) grammar when I \_\_\_\_\_ (fall) asleep.
- 11.- The scientists \_\_\_\_\_ (work) in their laboratory when they \_\_\_\_\_ (discover) the new drug.
- 12.- We \_\_\_\_\_ (climb) the wall when the gardener \_\_\_\_\_ (see) us.
- 13.- Vicky \_\_\_\_\_ (have) a beautiful dream when the alarm clock \_\_\_\_\_ (ring).
- 14.- As he \_\_\_\_\_ (run) for the bus he \_\_\_\_\_ (collide) with a street lamp.
- 15.- When he \_\_\_\_\_ (carry) a suitcase, he \_\_\_\_\_ (drop) it on his foot.
- 16.- \_\_\_\_\_ (you / do) your homework on the bus while you \_\_\_\_\_ (come) to school?
- 17.- The students \_\_\_\_\_ (read) the article when the last class \_\_\_\_\_ (finish).
- 18.- When I \_\_\_\_\_ (arrive) at the cinema, my friends \_\_\_\_\_ (wait) for me.
- 19.- \_\_\_\_\_ (you / listen) to the teacher when she \_\_\_\_\_ (explain) this exercise?
- 20.- Susan \_\_\_\_\_ (play) the piano while Mary \_\_\_\_\_ (sing).



# Esercizi Di Inglese Past Simple E Past Continuous

**Ying-Ying Zheng**



**Esercizi Di Inglese Past Simple E Past Continuous:**

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web write down the equilibrium equation you are investigating using the information provided by the introduction within the virtual lab activity equilibrium reaction  $\text{CO} + \text{H}_2\text{O} \rightleftharpoons \text{CO}_2 + \text{H}_2$  select each control test tube to compare the concentrations of

chemical equilibrium khan academy - Oct 22 2022

web quiz unit test introduction to equilibrium learn dynamic equilibrium direction of reversible reactions types of equilibrium constants learn writing equilibrium constant and reaction quotient expressions the equilibrium constant  $K$  calculating equilibrium constant  $K_p$  using partial pressures worked examples calculating equilibrium constants practice

**chemistry 1202 lechatelier s principle keq georgia public** - Nov 22 2022

web the study of reaction rates will lead you into the study of chemical equilibrium in this semester you will also study electrochemistry from batteries to electroplating and you will learn about acids bases and salts unit 15 is a brief study of the atomic nucleus which your teacher may introduce at any time during the year

**general chemistry ii lab 7 determination of the equilibrium** - Apr 27 2023

web equilibrium moles  $\text{HSCN}$   $2.00 \times 10^{-5}$  mol  $\text{Fe}^{3+}$   $3.00 \times 10^{-6}$  mol  $1.70 \times 10^{-5}$  mol  $\text{HSCN}$  knowing the number of moles of  $\text{Fe}^{3+}$  and  $\text{HSCN}$  present in the equilibrium mixture and the volume of the mixture we can easily find the concentrations of those two species mol  $\text{Fe}^{3+}$   $3.170 \times 10^{-5}$  mol  $\text{Fe}^{3+}$   $3.850 \times 10^{-4}$  M

equilibria lab report equilibria lab report introduction studocu - Jun 29 2023

web front matter 202 lab report assignment 1 2019 answers lab 1 lab 1 verification of gas laws equilibrium observation lab

**ap chem unit 7 overview equilibrium ap chemistry fiveable** - Feb 11 2022

web jan 29 2023 7 1 introduction to equilibrium let s start this unit out with a broaaaad overview of what equilibrium is all about in chemistry we define equilibrium as the point at which the rate of the forward reaction is equal to the rate of the reverse reaction we learned about rates in unit five as how quickly a reaction proceeds

*equilibrium questions practice khan academy* - Aug 20 2022

web equilibrium questions a covalent bond has a standard gibbs free energy of 400 kJ mol while the hydrogen bond and van der waals forces have substantially lower values respectively 25 and 5 kJ mol given such g values which of the following is equivalent to the ratio of the  $K_{eq}$  value of a covalent bond to that of a hydrogen bond

**chemical equilibrium chemistry library science khan academy** - Jan 25 2023

web unit 9 chemical bonds unit 10 gases and kinetic molecular theory unit 11 states of matter and intermolecular forces unit 12 chemical equilibrium unit 13 acids and bases unit 14 buffers titrations and solubility equilibria unit 15 thermodynamics

unit 16 redox reactions and electrochemistry

*equilibrium ap college chemistry science khan academy* - May 17 2022

web this unit explores the how and why of chemical equilibrium learn about reversible reactions the equilibrium constant le châtelier s principle solubility equilibria and more practice what you ve learned and study for the ap chemistry exam with more than 80 ap aligned questions

8 1 an introduction to equilibrium chemistry libretxts - Dec 24 2022

web jun 16 2020 summary skills to develop to understand what is meant by chemical equilibrium two methods to describe the likely result of a chemical reaction in the previous chapter we discussed the principles of chemical kinetics which deal with the rate of change or how quickly a given chemical reaction occurs

*11 1 introduction to chemical equilibrium chemistry libretxts* - Sep 01 2023

web nov 13 2022 define the equilibrium state of a chemical reaction system what is its practical significance state the meaning and significance of the following terms reversible reaction quantitative reaction kinetically inhibited reaction explain the meaning of the statement equilibrium is macroscopically static but microscopically dynamic very

**4 1 introduction to chemical equilibrium** - Jul 19 2022

web in a chemical equilibrium the forward and reverse reactions occur at equal rates and the concentrations of products and reactants remain constant if we run a reaction in a closed system so that the products cannot escape we often find the reaction does not give a 100 yield of products