

William Shakespeare Scavenger Hunt

Use books and the internet to find the answers to the following questions.



When is William Shakespeare Day?	Name Shakespeare's three children.	What was Shakespeare born?	Name the period of time Shakespeare was alive.
Which Queen watched some of Shakespeare's plays?	Name the Shakespearean play about a Scottish king.	What was the name of Shakespeare's son who died?	What was Shakespeare's wife called?
Which of Shakespeare's plays is a comedy about fairies and humans?	Where is the modern Globe Theatre?	Name Shakespeare's last play.	In which year did the original Globe Theatre burn down?
What actress is William Shakespeare's only daughter?	What is a ball play called?	Which festival are all Shakespearean plays based on?	Which of Shakespeare's plays is called 'The Merry Wives of Windsor'?
What words were used to describe Shakespeare as a poet?	Describe and label a ballad as shown in Shakespeare's play.	When did the first Globe Theatre open?	How old was Shakespeare when he died?



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web we then changed the volume between 5ml and 20ml and recorded the pressure at each this part of the lab showed that volume and pressure are inversely proportionate as the pressure would decrease if the volume was increased part b of the experiment looked at the relationship between temperature and pressure

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web procedure 1 the relationship between pressure and volume set the plunger of the syringe to 1 10 the capacity of the syringe the 2 0 ml mark if using a 20 ml syringe the volume is read from the edge of the bottom rib of the plunger attach the

syringe to the connector at the end of sensor

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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1 273 and concluded that the pressure would approach zero at 1 273 degrees c

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414l 22 414 l we can substitute 101 325kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature into the ideal gas equation and solve for r r

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web chemists sometimes make comparisons against a standard temperature and pressure stp for reporting properties of gases 273 15 k and 1 atm 101 325 kpa 1 at stp one mole of an ideal gas has a volume of about 22 4 l this is referred to as the standard molar volume figure 9 18

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