



Student Exploration: Balancing Chemical Equations

(This document is intended for printing.)

Essential Question: How do you balance a chemical equation? What are the coefficients in a balanced chemical equation? How do you balance a chemical equation?

What You Will Do: You will balance chemical equations and determine the coefficients.

1. **Balance** the chemical equation for the reaction of hydrogen gas and oxygen gas to form water. How many molecules of hydrogen gas and oxygen gas are needed to form 10 molecules of water? How many molecules of hydrogen gas and oxygen gas are needed to form 100 molecules of water?
2. **Balance** the chemical equation for the reaction of hydrogen gas and oxygen gas to form water. How many molecules of hydrogen gas and oxygen gas are needed to form 10 molecules of water? How many molecules of hydrogen gas and oxygen gas are needed to form 100 molecules of water?

What You Will Do

Use the following information to balance the chemical equation for the reaction of hydrogen gas and oxygen gas to form water.

Chemical Equation: $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$

The chemical equation shows that two molecules of hydrogen gas and one molecule of oxygen gas react to form two molecules of water.

Chemical Equation: $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$

How many molecules of hydrogen gas and oxygen gas are needed to form 10 molecules of water?

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What You Will Do



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AW Rasmussen



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