

Assessment

Chapter Test A

Chapter: Acids and Bases

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. Which of the properties listed below is *not* characteristic of an acid?
- a sour taste
 - a slippery feel
 - the ability to conduct an electric current
 - reactivity with metals
- _____ 2. Which is an example of a weak acid?
- HCl
 - H_2SO_4
 - H_3PO_4
 - HNO_3
- _____ 3. Acetic acid is found in
- vinegar.
 - the stomach.
 - antacids.
 - oranges.
- _____ 4. A characteristic of an Arrhenius base is that it
- is an electrolyte.
 - tastes bitter.
 - dissociates to form OH^- ions.
 - All of the above
- _____ 5. Which of the following is a Brønsted-Lowry base?
- HCl
 - HCO_3^-
 - H_3O^+
 - H_3PO_4
- _____ 6. In the reaction represented by the equation
- $$\text{H}_2\text{C}_2\text{O}_4(\text{aq}) + \text{CH}_3\text{NH}_2(\text{aq}) \rightleftharpoons \text{HC}_2\text{O}_4^-(\text{aq}) + \text{CH}_3\text{NH}_3^+(\text{aq}),$$
- which of these is a conjugate acid-base pair?
- $\text{H}_2\text{C}_2\text{O}_4(\text{aq})$ and $\text{CH}_3\text{NH}_2(\text{aq})$
 - $\text{H}_2\text{C}_2\text{O}_4(\text{aq})$ and $\text{CH}_3\text{NH}_3^+(\text{aq})$
 - $\text{CH}_3\text{NH}_2(\text{aq})$ and $\text{CH}_3\text{NH}_3^+(\text{aq})$
 - $\text{HC}_2\text{O}_4^-(\text{aq})$ and $\text{CH}_3\text{NH}_3^+(\text{aq})$

Modern Chemistry Chapter 7 Test Answers

Eleanor E. Hawkins



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Essentials of Modern Chemistry Charles Elwood Dull, 1918

A Complete Introduction to Modern NMR Spectroscopy Roger S. Macomber, 1997-12-23

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Books in Print Supplement, 2002

Lab Experiments Modern Chemistry Harold Clark Metcalfe, 1982

Instructor's Manual to Accompany Chemistry in the Modern World, Concepts and Applications Frank L. Wiseman, 1985

Boiler Technician 3 & 2 Phillip D. May, 1983

Study Guide Joan I. Senyk, Darrell D. Ebbing, 1984

Modern Inorganic Chemistry Joseph William Mellor, 1917

Basic Concepts of Chemistry, Study Guide Leo J. Malone, 2000

The text's three main goals are to introduce chemistry as a living, relevant science; to encourage learning and critical thinking; and to help readers overcome the math difficulties that impede their progress in chemistry. Designed to help readers master the principles of general chemistry, as a prep book it promotes active involvement with the material. There are special features throughout that reinforce concepts and help to develop strong problem-solving and study skills. Updated to include an Interactive Learning Ware problems CD containing several of the chapter-ending problems from the book in an interactive tutorial with feedback to help readers set up and solve problems.

Children's Books in Print R. R. Bowker Publishing, Bowker, 1999-12

Basic Chemical Principles Donada Peters, Edward I. Peters, 1988

Modern painters.-v.5-6. The stones of Venice.-v.7. Seven lamps of architecture. Lectures on architecture and painting. The study of architecture. Poetry of architecture.-v.8. Two paths ... on art. Lectures on art. Political economy of art. Pre-Raphaelitism. Notes on the construction of sheepfolds. King of the golden river.-v.9. Elements of drawing. Elements of perspective. Aratra pentelici.-v.10. Ariadne Florentina. Fors clavigera.-v.11. Sesame and lilies. Ethics of the dust. Crown of wild olive. Queen of the air.-v.12. Time and tide. Unto this last. Munera pulveris. Eagle's nest John Ruskin, 1885

Modern Physical Science

George R. Tracy,1983 **El-Hi Textbooks and Serials in Print, 1995** R R Bowker Publishing,1995-02 The United States Catalog Supplement, January 1918-June 1921 Eleanor E. Hawkins,1921 *The United States Catalog; Books in Print January 1, 1912* Marion Effie Potter,1921 **Publishers' Circular and Booksellers' Record of British and Foreign Literature** ,1898 *Safe Living, Instructor's Guide* ,1960 Digest ,1891

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