

Standard Enthalpy of Formation (ΔH°_f) at 298 K

Formula	KJ/mol	Formula	KJ/mol	Formula	KJ/mol
Carbon		Chlorine		Sodium	
C (s, graphite)	0	Cl (g)	121.3	Na (s)	0
C (s, diamond)	1.88	Cl ₂ (g)	0	Na (g)	107.5
CH ₄ (g)	-74.6	HCl (g)	-92.3	NaCl (s)	-411.2
CH ₃ OH (l)	-238.6	Fluorine		Na ₂ CO ₃ (s)	-1130.7
C ₂ H ₂ (g)	227.4	F (g)	79.38	NaHCO ₃ (s)	-950.8
C ₂ H ₄ (g)	52.4	F ₂ (g)	0	Sulfur	
C ₂ H ₆ (g)	-84.68	HF (g)	-273.3	S ₈ (s, rhombic)	0
C ₂ H ₅ OH (l)	-277.6	Nitrogen		S ₈ (s, monoclinic)	0.3
C ₃ H ₈ (g)	-103.85	N ₂ (g)	0	SO ₂ (g)	-296.8
C ₃ H ₆ O (l, acetone)	-248.4	NH ₃ (g)	-45.9	SO ₃ (g)	-395.7
Hydrogen		NH ₄ (s)	-365.6	H ₂ SO ₄ (l)	-814.0

Standard Enthalpy Of Formation For Various Compounds

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Standard Enthalpy Of Formation For Various Compounds:

Unveiling the Power of Verbal Beauty: An Mental Sojourn through **Standard Enthalpy Of Formation For Various Compounds**

In a global inundated with monitors and the cacophony of quick transmission, the profound power and psychological resonance of verbal art frequently fade in to obscurity, eclipsed by the regular onslaught of sound and distractions. Yet, situated within the lyrical pages of **Standard Enthalpy Of Formation For Various Compounds**, a charming function of fictional beauty that pulses with natural feelings, lies an memorable trip waiting to be embarked upon. Composed with a virtuoso wordsmith, that magical opus manuals viewers on an emotional odyssey, gently exposing the latent possible and profound impact stuck within the elaborate internet of language. Within the heart-wrenching expanse with this evocative examination, we will embark upon an introspective exploration of the book is central subjects, dissect their captivating publishing model, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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