

## Chemistry Lab Report: Identifying an Unknown Compound

In this lab, I identified an unknown compound by testing and comparing the experimental results of the compound with the known compounds. The unknown compound in this lab was a white solid, soluble in water, and gave a positive test for the presence of the carbonate ion. The unknown compound was identified as sodium carbonate,  $\text{Na}_2\text{CO}_3$ . The unknown compound was a white solid, soluble in water, and gave a positive test for the presence of the carbonate ion. The unknown compound was identified as sodium carbonate,  $\text{Na}_2\text{CO}_3$ . The unknown compound was a white solid, soluble in water, and gave a positive test for the presence of the carbonate ion. The unknown compound was identified as sodium carbonate,  $\text{Na}_2\text{CO}_3$ .

The unknown compound and the three known compounds were all tested with water and hydrochloric acid. The unknown compound was a white solid, soluble in water, and gave a positive test for the presence of the carbonate ion. The unknown compound was identified as sodium carbonate,  $\text{Na}_2\text{CO}_3$ . The unknown compound was a white solid, soluble in water, and gave a positive test for the presence of the carbonate ion. The unknown compound was identified as sodium carbonate,  $\text{Na}_2\text{CO}_3$ .

When the unknown compound was tested with water, it dissolved completely. The unknown compound was a white solid, soluble in water, and gave a positive test for the presence of the carbonate ion. The unknown compound was identified as sodium carbonate,  $\text{Na}_2\text{CO}_3$ . The unknown compound was a white solid, soluble in water, and gave a positive test for the presence of the carbonate ion. The unknown compound was identified as sodium carbonate,  $\text{Na}_2\text{CO}_3$ . The unknown compound was a white solid, soluble in water, and gave a positive test for the presence of the carbonate ion. The unknown compound was identified as sodium carbonate,  $\text{Na}_2\text{CO}_3$ .

The unknown compound was tested with hydrochloric acid. The unknown compound was a white solid, soluble in water, and gave a positive test for the presence of the carbonate ion. The unknown compound was identified as sodium carbonate,  $\text{Na}_2\text{CO}_3$ . The unknown compound was a white solid, soluble in water, and gave a positive test for the presence of the carbonate ion. The unknown compound was identified as sodium carbonate,  $\text{Na}_2\text{CO}_3$ .

# Identifying Unknown Solutions Chemistry Lab Report

**Harold Clark Metcalfe**



## Identifying Unknown Solutions Chemistry Lab Report:

**Chemistry in the Laboratory** James M. Postma, Julian L. Robert, J. Leland Hollenberg, 2004-03-12 This clearly written class tested manual has long given students hands on experience covering all the essential topics in general chemistry Stand alone experiments provide all the background introduction necessary to work with any general chemistry text This revised edition offers new experiments and expanded information on applications to real world situations

**Laboratory Manual for Principles of General Chemistry** J. A. Beran, Mark Lassiter, 2022-08-16 Laboratory Manual for Principles of General Chemistry 11th Edition covers two semesters of a general chemistry laboratory program The material focuses on the lab experiences that reinforce the concepts that not all experimental conclusions are the same and depend on identifying an appropriate experimental procedure selecting the proper apparatus employing the proper techniques systematically analyzing and interpreting the data and minimizing inherent variables As a result of good data a scientific and analytical conclusion is made which may or may not be right but is certainly consistent with the data Experiments write textbooks textbooks don t write experiments A student s scientific literacy grows when experiences and observations associated with the scientific method are encountered Further experimentation provides additional cause effect observations leading to an even better understanding of the experiment The 11th edition s experiments are informative and challenging while offering a solid foundation for technique safety and experimental procedure The reporting and analysis of the data and the pre and post lab questions focus on the intuitiveness of the experiment The experiments may accompany any general chemistry textbook and are compiled at the beginning of each curricular unit An Additional Notes column is included in each experiment s Report Sheet to provide a space for recording observations and data during the experiment Continued emphasis on handling data is supported by the Data Analysis section

**Laboratory Manual for Principles of General Chemistry** Jo Allan Beran, 2010-11-01 This new edition of the Beran lab manual emphasizes chemical principles as well as techniques The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures

**Inquiry-based Experiments in Chemistry** Valerie Ludwig Lechtanski, 2000 Inquiry Based Experiments in Chemistry is an alternative to those cookbook style lab manuals providing a more accurate and realistic experience of scientific investigation and thought for the high school chemistry or physical science student

Working with Chemistry Donald J. Wink, Sharon Fetzer-Gislason, Julie Ellefson Kuehn, 2004-02-20 With this modular laboratory program students build skills using important chemical concepts and techniques to the point where they are able to design a solution to a scenario drawn from a professional environment The scenarios are drawn from the lives of people who work with chemistry every day ranging from field ecologists to chemical engineers and include many health professionals as well

*Basic Laboratory Principles in General Chemistry* Fitzgerald B. Bramwell, 1990 *Merrill*

*Chemistry-Lab.Manual* Smoot,1994-07     Scientific and Technical Aerospace Reports ,1963     *Addison-Wesley Chemistry Laboratory Manual* Dennis D. Staley,1995     **Prentice Hall Chemistry** Antony C. Wilbraham,2008     *Lab Experiments for Modern Chemistry* Tzimopoulo,1990     *General, Organic, and Biological Chemistry Study Guide and Selected Solutions* Karen C. Timberlake,2001-11 Keyed to the learning goals in the text this guide is designed to promote active learning through a variety of exercises with answers and mastery exams The guide also contains complete solutions to odd numbered problems     *Lab Experiments Modern Chemistry* Harold Clark Metcalfe,1982     **Laboratory Manual for Chemistry** Karen C. Timberlake,1976     **Exercises in General, Organic, and Biological Chemistry** Arne Nels Langsjoen,1973     *Experiences in Cooperative Learning* Susan C. Nurrenbern,1995 Contains many examples of activities ranging from science at the middle school level to college undergraduate chemistry course     *General Chemistry* John F. Baxter,1956

**Laboratory Inquiry in Chemistry** Richard C. Bauer,Richard Bauer,James P. Birk,Douglas J. Sawyer,2005 LABORATORY INQUIRY IN CHEMISTRY Second Edition provides a unique set of guided inquiry investigations that focus on constructing knowledge about the conceptual basis of laboratory techniques instead of simply learning techniques By focusing on developing skills for designing experiments solving problems thinking critically and selecting and applying appropriate techniques the authors expose students to a realistic laboratory experience typical of the practicing chemist The Second Edition features six new experiments and is accompanied by a revised and updated Instructor s Manual available online This new edition continues the proven three phase learning cycle exploration of chemical behaviors within the context of the problems posed concept invention the use of data and observations to construct accepted scientific knowledge about the concepts explored in the laboratory investigation and concept application where students apply their conceptual understanding of the investigation at hand by modifying or extending the experiments and write a report that emphasizes conceptual relevance These college and honors level inquiry based experiments correlate well with the recommended experiments outlined by the Advanced Placement Chemistry Development Committee     **Chemistry** Karen Timberlake,1999 Suitable for one or two term lab courses covering general organic and biological chemistry this new edition written by Karen Timberlake features many improvements to the insightful experiments that have made it the leading lab manual Each experiment encourages critical thinking with laboratory goals discussion of related concepts clear instructions new pre lab questions and comprehensive report pages Forty one experiments illustrate the basic principles of chemistry     *Virtual Chemlab* Brian F. Woodfield,Matthew C. Asplund,Steven Haderlie,2006 Contains a full virtual lab environment as well as the pre arranged labs that are refer enced in the workbook and at the end of the chapter in the textbook Virtual ChemLab can be run directly from the CD or installed on the student s computer

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