

Cambridge IGCSE™ Chemistry

Workbook answers

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1 States of matter

Core

- 1 In ice, I (the water molecule) am bonded to other water molecules and vibrating in a fixed position. As the temperature rises, there is enough energy for some of the bonds between us to weaken and begin to break due to the increase in vibrations. This allows me to move further from my neighbouring molecules and I am able to move past them. As the temperature continues to rise, more of the bonds between us begin to break and I move away from my neighbouring molecules and move much more quickly. At 100 °C I have so little attraction to my neighbouring molecules that I am able to break away and move a long way away from them.
- 2
 - a
 - i When a substance changes from a solid to a liquid
 - ii When substances are changed into different substances by chemical reaction
 - iii When a gas changes into a liquid
 - iv When a liquid changes into a gas
 - v When a liquid changes into a solid
 - vi When heating of a liquid causes it to vaporize
 - b
 - i Condensation
 - ii Melting
- 3
 - a Diffusion cannot occur in solids because the particles are bonded to one another more strongly than those in liquids and gases, which do diffuse. This prevents them from being able to move relative to one another.
 - b Gases can be compressed more than liquids because the particles in the gas are much more spread out than those in a liquid. This means that the gas particles can be squeezed together more than those in a liquid, which are much closer to one another.
 - c Liquids can flow because their particles have much weaker bonds between them than those in a solid. This means that the particles in a liquid can move over one another.

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Karin Nielsen-Saines



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