

রাজশাহী বোর্ড

বহুনির্বাচনি অধীকার
 ক্লাস/সাল
 বিত্তীয় পর
 [২০২০ সালের নিবেদন অনুযায়ী]
 সময়—২৫ মিনিট
 পূর্ণমান—২০

বিষয় কোড :

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 সেট :

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বিবেচন গ্রহণ করুন : সরবরাহকৃত বহুনির্বাচনি অধীকার উত্তরগুলোতে কোনো ভ্রমের ক্ষমিক দৃষ্টান্তে প্রদত্ত বহুনির্বাচনিত সূত্রসমূহ হতে সঠিক/সর্বোৎকৃষ্ট উত্তরের সূত্রটি বলা পর্যন্তে কলাম দ্বারা সম্পূর্ণ করাটি করুন। প্রতিটি প্রশ্নের মান ১।

প্রশ্নগুলো কোনো প্রকার দাপ/সিক্সে লেখা যাবে না।

- অ্যামাইডের কার্বকটী সূত্র কোনটি?
 (ক) - CNS
 (খ) - CO - NH -
 (গ) - NH₂
 (ঘ) - CO - NH₂ *
- C₄H₈ বৌদটিতে কয়টি গ্রুপের বন্ধন আছে?
 (ক) 12
 (খ) 10
 (গ) 11 *
 (ঘ) 9
- 500 mL 0.05M Na₂CO₃ দ্রবণে কত গ্রাম Na₂CO₃ থাকবে?
 (ক) 2.65 *
 (খ) 6.30
 (গ) 5.30
 (ঘ) 10.60
- কোনটির আয়ন বিদ্যুত সর্বোচ্চ?
 (ক) কপার
 (খ) ব্রহ্মিয়ডেন
 (গ) সোডিয়াম *
 (ঘ) সিলিকন
- কোনটি সবচেয়ে বেশি পটাসিয়াম বিকারক?
 (ক) Li *
 (খ) K
 (গ) Na
 (ঘ) Ag
- কোন অক্সিডার্ট আয়ন প্রতিকার নির্দেশ করে?
 (ক) Zn²⁺/Zn
 (খ) Zn/Zn²⁺ *
 (গ) Cu²⁺/Cu
 (ঘ) H⁺/H₂, Pt
- বায়ুয় গ্যাসের সমীকরণ কোনটি?
 (ক) PV = nRT
 (খ) PV = RT
 (গ) $(P + \frac{n^2a}{V^2})(V - nb) = nRT$ *
 (ঘ) PV = $\frac{1}{3} nN_c^2$
- SATP তে 2 মোল O₂ গ্যাসের আয়তন কত?
 (ক) 22.789 L
 (খ) 24.789 L
 (গ) 45.578 L
 (ঘ) 49.578 L *

নিম্নের উদাহরণটি পড়ুন এবং পরবর্তী দুইটি প্রশ্নের উত্তর দিন।



Y - গ্যাসের পরমাণুর সংখ্যের পটাসিয়ামের ইলেকট্রন বিশাল na^2np^4 , যেখানে $n = 2$

- STP তে 0.25 mol Y গ্যাসের আয়তন কত?
 (ক) 3.5g
 (খ) 8g *
 (গ) 7g
 (ঘ) 16g
- উদাহরণের X - গ্যাস—
 I. এর সোডিয়াম আয়ন $x = 1$
 II. PV = nRT সূত্র বোলে চলে
 III. এর আয়তন বোলে মনো অক্সিজেন বলা বিদ্যমান
 কোনটি সঠিক?
 (ক) I ও II *
 (খ) II ও III
 (গ) I, II ও III
- C(CH₃)₂-OH বৌদটির IUPAC নাম কী?
 (ক) 2-মিথাইল প্রোপেন-2-অল
 (খ) অইলো মিটাইল অ্যালকোহল
 (গ) 2, 2-ডাইমিথাইল ইথানল *
 (ঘ) মিটাইল
- মিথাইল অক্সিজেনের সর্ব পটাসিয়ামের pH সীমা কোনটি?
 (ক) 3.2 - 4.2 *
 (খ) 4.3 - 5.4
 (গ) 5.0 - 8.0
 (ঘ) 8.2 - 10.0
- কোনটি HCO₃⁻ এর অনুবর্তী কারক?
 (ক) H₂CO₃
 (খ) CO₂
 (গ) H₂O
 (ঘ) CO₃²⁻ *

[পর পৃষ্ঠা প্রদর্শন]

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