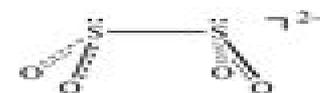


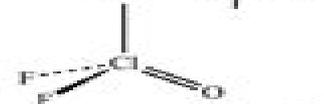
- e.  $S_2O_8^{2-}$  has  $SO_2$  units with an angle of about  $30^\circ$  between their planes, in an eclipsed conformation.



- f.  $N_2H_4$  has a trigonal pyramidal shape at each N, and a *gauche* conformation. There is one lone pair on each N.

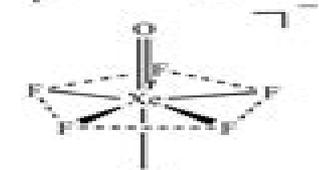


- g.  $ClOF_2^+$  is a distorted trigonal pyramid with one lone pair and double bond character in the Cl—O bond.



- h.  $CS_2$ , like  $CO_2$ , is linear with double bonds.  $S=C=S$

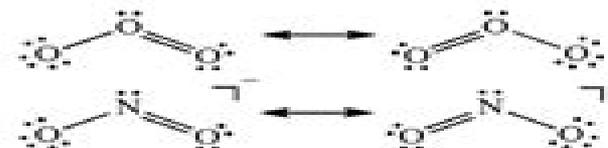
- i. The structure of  $XeOF_5^-$  is based on a pentagonal bipyramid, with a lone pair and the oxygen atom in axial positions. See K. O. Christie et al., *Inorg. Chem.*, **1995**, *34*, 1868 for evidence in support of this structure.



- 3.11** All the halate ions are trigonal pyramids; as the central atom increases in size, the bonding pairs are farther from the center, and the lone pair forces a smaller angle. The decreasing electronegativity  $Cl > Br > I$  of the central atom also allows the electrons to be pulled farther out, reducing the *bp-bp* repulsion.

- 3.12** a.  $AsH_3$  should have the smallest angle, since it has the largest central atom. This minimizes the bond pair—bond pair repulsions and allows a smaller angle. Arsenic is also the least electronegative central atom, allowing the electrons to be drawn out farther and lowering the repulsions further. Actual angles:  $AsH_3 = 91.8^\circ$ ,  $PH_3 = 93.8^\circ$ ,  $NH_3 = 106.6^\circ$ .
- b. Cl is larger than F, and F is more electronegative and should pull the electrons farther from the S, so the F—S—F angle should be smaller in  $OSF_2$ . This is consistent with the experimental data: the F—S—F angle in  $OSF_2$  is  $92.3^\circ$  and the Cl—S—Cl angle in  $OSCl_2$  is  $96.2^\circ$ .

- c.  $NO_2^-$  has rather variable angles ( $115^\circ$  and  $132^\circ$ ) in different salts. The sodium salt ( $115.4^\circ$ ) has a slightly smaller angle than  $O_3$  ( $116.8^\circ$ ). The N—O electronegativity difference should pull electrons away from N, reducing the *bp-bp* repulsion and the angle.



- d.  $BrO_3^-$  ( $104^\circ$ ) has a slightly smaller angle than  $ClO_3^-$  ( $107^\circ$ ), since it has a larger central atom. In addition, the greater electronegativity of Cl holds the electrons closer and increases *bp-bp* repulsion.

- 3.13** a.  $N_3^-$  is linear, with two double bonds.  $O_3$  is bent (see solution to 3.12.c), with one double bond and a lone pair on the central O caused by the extra pair of electrons.

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web oct 12 2022 hiraeth zamanı ve mekanı aşan bir özlemdir hiraeth doğrudan İngilizce çevirisi olmayan galce bir kelimedir hiraeth yurt özlemi yuvaya ya da vatana hasret anlamına gelir genellikle tam yeri bilinmeyen veya hatta kaybolmuş olabilecek bir yere derin bir duygusal bağlılık olarak tanımlanır