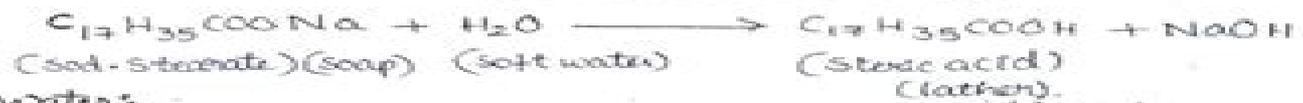


1. Water Technology.



* Water:

→ It is an important natural source and occupies $\frac{3}{4}$ th of earth surface. Water is essential for existence of life as most of the compounds are soluble in it.

→ Hence, water is treated as a universal solvent.

In addition to drinking purpose, water is also used in Industries, for production of chemicals, textiles, steel, paints, etc.... and also used in generation of electricity.

* Types of water:

Water is classified into two types based on action with soap.

→ Soft water: The water which produces lather immediately and easily with soap ($\text{C}_{17}\text{H}_{35}\text{COONa}$) is called soft water.



→ Hard water: The water which does not produce lather (or) a very little amount of lather with soap is called Hard water. It produces white precipitate with soap due to formation of Calcium and Magnesium salts forms complex with the soap.



* Expression of hardness (or) degree of hardness:

The total amount of hardness producing salts which are present in the definite quantity of water is called degree of hardness (or) expression of hardness.

The degree of hardness is expressed in terms of CaCO_3 equivalents because CaCO_3 M.Wt. is "100gm" which is easy to calculate and is insoluble in water.

$$\text{CaCO}_3 \text{ equivalents} = \frac{\text{Wt. of Hardness causing salt}}{\text{M.Wt. of H.C.S.}} \times 100.$$

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