

Chemistry Periodic Table Test Chapter 6 Chem IH

Matching Match each item with the correct statement below.

- | | |
|----------------------|---------------------|
| a. electronegativity | e. transition metal |
| b. ionization energy | f. periodic law |
| c. electrons | g. group |
| d. metal | h. period |

- _____ 1. vertical column in the periodic table
_____ 2. type of element characterized by the presence of electrons in the *d* orbital
_____ 3. type of element characterized by losing electrons easily
_____ 4. ability of an atom to attract electrons when the atom is in a compound
_____ 5. energy required to remove an electron from an atom

Multiple Choice

Identify the letter of the choice that best completes the statement or answers the question.

- _____ 6. What is another name for the main group elements?
a. s block elements
b. p block elements elements
c. s & p block elements
d. d block elements
- _____ 7. What is another name for the transition metals?
a. noble gases
b. s block elements elements
c. p block elements elements
d. d block elements
- _____ 8. Who arranged the elements according to atomic mass and used the arrangement to predict the properties of missing elements?
a. Henry Moseley
b. Antoine Lavoisier
c. John Dalton
d. Dmitri Mendeleev
- _____ 9. Which of the following categories includes the majority of the elements?
a. metalloids
b. liquids
c. metals
d. nonmetals
- _____ 10. In which of the following sets is the symbol of the element, the number of protons, and the number of electrons given correctly?
a. In, 49 protons, 49 electrons
b. Zn, 30 protons, 60 electrons
c. Cs, 55 protons, 132.9 electrons
d. F, 19 protons, 19 electrons
- _____ 11. The atomic number of an element is the total number of which particles in the nucleus?
a. neutrons
b. protons
c. electrons
d. protons and electrons
- _____ 12. What element has the electron configuration $1s^2 2s^2 2p^6 3s^2 3p^2$?
a. nitrogen
b. selenium
c. silicon
d. silver
- _____ 13. Which subatomic particle plays the greatest part in determining the properties of an element?
a. proton
b. electron
c. neutron
d. none of the above
- _____ 14. Which of the following elements is a transition metal?
a. cesium
b. copper
c. tellurium
d. tin

Periodic Table Chapter Test A Answer Key

B Lingard



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