

Chemistry 12 - Review of Significant Digits

TOTAL 28

The rules for zeros in significant digits are as follows:

a) All zeros between non-zero digits are significant.

b) Zeros at the beginning of a number (eg. 0.0095) are NOT SIGNIFICANT!

If the number 0.0095 was written in scientific notation, it would be: 9.5×10^{-3} . The exponent is not counted as significant so this number has 2 significant digits.

c) Zeros on the right side of a number (at the end) are significant if the DECIMAL POINT is shown.

eg) 50.00 has 4 significant digits
43.0 has 3 significant digits
20. has 2 significant digits
100. has 3 significant digits

d) Zeros to the left of an UNDERSTOOD decimal point are NOT significant.

eg) 300 has 1 significant digit
10 000 has 1 significant digit
12 320 has 4 significant digits
420 has 2 significant digits

1. Find the number of significant digits in each of the following measurements:

- | | | | |
|--------------------------------|----------|---------------------------------|----------|
| a) 3.4005 | <u>5</u> | f) 9.080×10^{-3} | <u>4</u> |
| b) 2580 | <u>3</u> | g) 1.00 | <u>3</u> |
| c) 3.20×10^{-2} | <u>3</u> | h) 0.0027890 | <u>5</u> |
| d) 0.000308 | <u>3</u> | i) 320 000 | <u>2</u> |
| e) 23.000 | <u>5</u> | j) 9 | <u>1</u> |

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