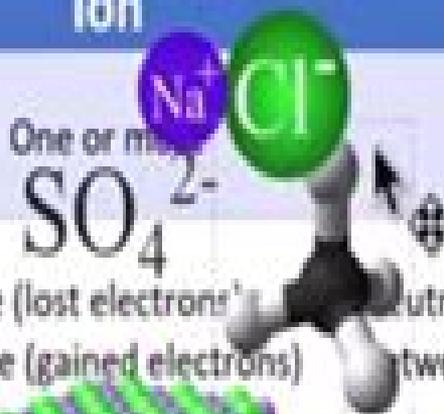


# Atoms, Ions, & Molecules

	Atom	Ion	Molecule
Number of particles	 One 	One or more 	Two or more 
Charge	Neutral (equal protons and electrons)	Positive (lost electrons) negative (gained electrons)	Neutral (electrons shared between atoms)
Formation	Cannot be further broken down without losing chemical identity.	Formed by sharing electrons	Formed by sharing electrons between atoms.
Examples	Helium (He), Oxygen (O), Iron (Fe)	Sodium ion (Na <sup>+</sup> ), Chloride ion (Cl <sup>-</sup> ), Sulfate ion (SO <sub>4</sub> <sup>2-</sup> )	Water (H <sub>2</sub> O), Carbon dioxide (CO <sub>2</sub> ), Sugar (C <sub>6</sub> H <sub>12</sub> O <sub>6</sub> )



# Chapter 7 Compounds Ions And Molecules

**H Kauffman**



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