

## IONIC BONDING

Ionic bonding occurs when a metal transfers one or more electrons to a nonmetal in an effort to obtain a stable octet of electrons. For example, the transfer of an electron from sodium to chlorine can be depicted by a Lewis dot diagram.



Calcium would need two chlorine atoms to get rid of its two valence electrons.



Show the transfer of electrons in the following combinations.

1. K + F



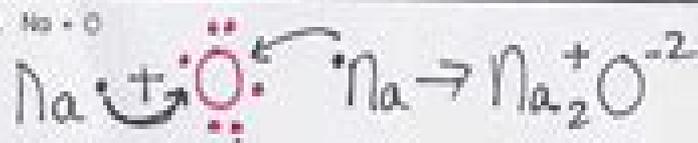
2. Mg + I



3. Be + S



4. Na + O



5. Al + Br



## COVALENT BONDING

Name \_\_\_\_\_

Covalent bonding occurs when two or more nonmetals share electrons, attempting to gain a stable octet of electrons at least part of the time. For example:

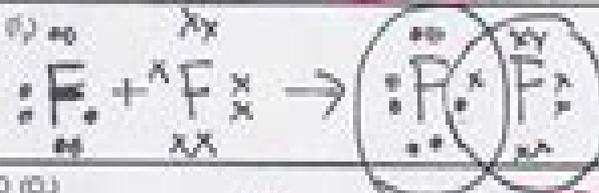


Show how covalent bonding occurs in each of the following pairs of atoms. Atoms may share one, two or three pairs of electrons.

1. H + H (H<sub>2</sub>)



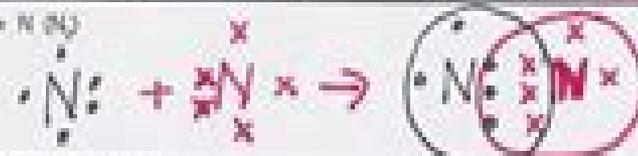
2. F + F (F<sub>2</sub>)



3. O + O (O<sub>2</sub>)



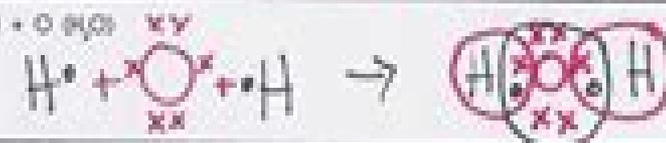
4. N + N (N<sub>2</sub>)



5. C + O (CO<sub>2</sub>)



6. H + O (H<sub>2</sub>O)



# Chapter 8 Covalent Bonding Worksheet Answers

**ML Yell**



## **Chapter 8 Covalent Bonding Worksheet Answers:**

## Reviewing **Chapter 8 Covalent Bonding Worksheet Answers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Chapter 8 Covalent Bonding Worksheet Answers**," an enthralling opus penned by a very acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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