



## The Chemistry of Life

## Chapter Test A

### Multiple Choice

Directions: Select the best answer for each of the questions by circling the number of the best possible answer.

- \_\_\_\_\_ 1. If an atom has a mass number of 23 and 11 electrons, the atom has a net charge of
  - a. 0
  - b. -11
  - c. 11
  - d. 23
- \_\_\_\_\_ 2. If an atom has a mass of 23 grams and 11 protons, the atom's net charge is
  - a. 0
  - b. -11
  - c. 11
  - d. 23
- \_\_\_\_\_ 3. Which of the following would be considered an acid?
  - a. one amino acid building block of a protein
  - b. one amino acid building block of a nucleic acid
  - c. one amino acid building block of a sugar
  - d. two amino acid building blocks of a protein
- \_\_\_\_\_ 4. A molecule that is neutral in the standard
  - a. bonding condition
  - b. bonding process
  - c. bonding structure
  - d. bonding process
- \_\_\_\_\_ 5. Which molecule can pair with the
  - a. sugar with being slightly positive and the phosphate with being slightly negative
  - b. sugar with being slightly positive
  - c. sugar with being slightly negative
  - d. sugar with being slightly negative and the phosphate with being slightly positive
- \_\_\_\_\_ 6. If you are left with half of a sugar, you are left with
  - a. positive and negative charges
  - b. positive and negative charges
  - c. positive and negative charges
  - d. positive only
- \_\_\_\_\_ 7. A substance with a pH of 7 is called
  - a. an acid
  - b. basic
  - c. neutral
  - d. a compound
- \_\_\_\_\_ 8. Nucleic acids are made from
  - a. proteins
  - b. lipids
  - c. acids
  - d. monomers
- \_\_\_\_\_ 9. Why is coffee considered compared to other beverages?
  - a. Coffee does not contain any acids and therefore has a different structure.
  - b. Coffee does not have a different taste.
  - c. With coffee, you are not interacting with sugars and building blocks.
  - d. With coffee, you are not interacting with amino acids and nucleic acids.

# Chapter Test Chemistry Of Life Answer Key

**William K. Purves**



## Chapter Test Chemistry Of Life Answer Key:

*Chapter Resource 2 Chemistry of Life Biology* Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004

Chemistry and Life John William Hill, Dorothy M. Feigl, Stuart J. Baum, 1993      Chemistry and Life John W. Hill, 1997

**Life** William K. Purves, 2001 Authoritative thorough and engaging Life The Science of Biology achieves an optimal balance of scholarship and teachability never losing sight of either the science or the student The first introductory text to present biological concepts through the research that revealed them Life covers the full range of topics with an integrated experimental focus that flows naturally from the narrative This approach helps to bring the drama of classic and cutting edge research to the classroom but always in the context of reinforcing core ideas and the innovative scientific thinking behind them Students will experience biology not just as a litany of facts or a highlight reel of experiments but as a rich coherent discipline

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