

name Key

DUE date/TEST date 10/24 - Mon 10/25 - Tue

# Practice Problems Unit 5 - Nuclear Chemistry

Directions: Answer each of the following questions. Show ALL work in the space provided with units and correct sig figs to receive credit.

## Practice Problems - Nuclear Reactions

1. Fill out the following chart to help predict types of radiation.

Type of Radiation	How can you recognize that problem?	
beta decay	$\frac{0}{-1}e$	mass # ↓ 0 atomic # ↓ 1
alpha decay	$\frac{4}{2}He$	mass # ↓ 4 atomic # ↓ 2
gamma decay		γ - no change

2. Fill out the chart based on your knowledge of nuclear reactions.

Element	Protons	Neutrons	Electrons	Stable?	Type of Radiation	Balanced Nuclear Reaction
carbon-14	6	8	6	no	beta decay	$^{14}_6C \rightarrow ^{14}_7N + ^0_{-1}\beta$
radium-226	88	$226 - 88 = 138$	88	no	alpha decay	$^{226}_{88}Ra \rightarrow ^4_2He + ^{222}_{86}Rn$
$^{65}Cu$	29	$65 - 29 = 36$	29	no	beta decay	$^{65}_{29}Cu \rightarrow ^0_{-1}e + ^{65}_{30}Zn$
$^{127m}I$	53	$127 - 53 = 74$	53	no	gamma decay	$^{127m}_{53}I \rightarrow ^0_0\gamma + ^{127}_{53}I$
$^{186m}Re$	75	$186 - 75 = 111$	75	no	gamma decay	$^{186m}_{75}Re \rightarrow ^0_0\gamma + ^{186}_{75}Re$
nobelium-259	102	$259 - 102 = 157$	102	no	alpha decay	$^{259}_{102}No \rightarrow ^4_2He + ^{255}_{100}Fm$

3. How are the mass number and atomic number of the nucleus affected by the following:

- Beta decay mass # same atomic # ↑ 1
- Alpha decay mass # ↓ 4 atomic # ↓ 2
- Gamma decay mass # same atomic # same

4. Complete and balance each of the following nuclear reactions.



5. Bromine -82 decays to produce Krypton - 82.

- What type of decay is this? justify your answer! beta decay - mass stayed the same + atomic # ↑ 1
- Write the nuclear reaction for this decay.  $^{82}_{35}Br \rightarrow ^{82}_{36}Kr + ^0_{-1}e$

6. Neptunium - 237 undergoes decay to Protactinium - 233.

- What type of decay is this? justify your answer! alpha decay → mass # ↓ 4 + atomic # ↓ 2
- Write the nuclear reaction for this decay.  $^{237}_{93}Np \rightarrow ^{233}_{91}Pa + ^4_2He$

7. Write a balanced nuclear reaction for the beta decay of:



8. Write a balanced nuclear reaction for the alpha decay of:



# Chapter 22 Nuclear Chemistry Section 1 Review Answers

**Kaplan Test Prep**



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## **Table of Contents Chapter 22 Nuclear Chemistry Section 1 Review Answers**

1. Understanding the eBook Chapter 22 Nuclear Chemistry Section 1 Review Answers
  - The Rise of Digital Reading Chapter 22 Nuclear Chemistry Section 1 Review Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 22 Nuclear Chemistry Section 1 Review Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chapter 22 Nuclear Chemistry Section 1 Review Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 22 Nuclear Chemistry Section 1 Review Answers
  - Personalized Recommendations
  - Chapter 22 Nuclear Chemistry Section 1 Review Answers User Reviews and Ratings

- Chapter 22 Nuclear Chemistry Section 1 Review Answers and Bestseller Lists
- 5. Accessing Chapter 22 Nuclear Chemistry Section 1 Review Answers Free and Paid eBooks
  - Chapter 22 Nuclear Chemistry Section 1 Review Answers Public Domain eBooks
  - Chapter 22 Nuclear Chemistry Section 1 Review Answers eBook Subscription Services
  - Chapter 22 Nuclear Chemistry Section 1 Review Answers Budget-Friendly Options
- 6. Navigating Chapter 22 Nuclear Chemistry Section 1 Review Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Chapter 22 Nuclear Chemistry Section 1 Review Answers Compatibility with Devices
  - Chapter 22 Nuclear Chemistry Section 1 Review Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chapter 22 Nuclear Chemistry Section 1 Review Answers
  - Highlighting and Note-Taking Chapter 22 Nuclear Chemistry Section 1 Review Answers
  - Interactive Elements Chapter 22 Nuclear Chemistry Section 1 Review Answers
- 8. Staying Engaged with Chapter 22 Nuclear Chemistry Section 1 Review Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chapter 22 Nuclear Chemistry Section 1 Review Answers
- 9. Balancing eBooks and Physical Books Chapter 22 Nuclear Chemistry Section 1 Review Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chapter 22 Nuclear Chemistry Section 1 Review Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Chapter 22 Nuclear Chemistry Section 1 Review Answers
  - Setting Reading Goals Chapter 22 Nuclear Chemistry Section 1 Review Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chapter 22 Nuclear Chemistry Section 1 Review Answers
  - Fact-Checking eBook Content of Chapter 22 Nuclear Chemistry Section 1 Review Answers
  - Distinguishing Credible Sources

13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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