

# COVALENT BONDING

Name \_\_\_\_\_

Covalent bonding occurs when two or more nonmetals share electrons, attempting to attain a stable octet of electrons at least part of the time. For example:

*H wants 2 electrons  
all other atoms 8*



*electrons in the middle share*

Show how covalent bonding occurs in each of the following pairs of atoms. Atoms may share one, two or three pairs of electrons.

*wanted available share*

<p>1. H + H (H<sub>2</sub>)</p> <p>W - 4 A - 2 S - 2</p>	<p><i>wanted 2 + 2 available</i></p> <p><math>\text{H} \cdot \cdot \text{H}</math></p> <p><math>\text{H} \cdot \quad \cdot \text{H}</math></p>	<p><math>\text{H} - \text{H}</math></p> <p>single bond = share 2 e<sup>-</sup></p>
<p>2. F + F (F<sub>2</sub>)</p> <p>W - 8 + 8 = 16 A - 7 + 7 = 14 S - 2</p>	<p><math>\cdot \ddot{\text{F}} \cdot \cdot \ddot{\text{F}} \cdot</math></p> <p><math>\cdot \ddot{\text{F}} \cdot \quad \cdot \ddot{\text{F}} \cdot</math></p>	<p><math>:\ddot{\text{F}} - \ddot{\text{F}}:</math></p> <p>single bond 2e<sup>-</sup></p>
<p>3. O + O (O<sub>2</sub>)</p> <p>W - 8 + 8 = 16 A - 6 + 6 = 12 S - 4</p>	<p><math>\cdot \ddot{\text{O}} \cdot \cdot \ddot{\text{O}} \cdot</math></p> <p><math>\cdot \ddot{\text{O}} \cdot \quad \cdot \ddot{\text{O}} \cdot</math></p>	<p><math>:\ddot{\text{O}} = \ddot{\text{O}}:</math></p> <p>4 double bond</p>
<p>4. N + N (N<sub>2</sub>)</p> <p>W - 8 + 8 = 16 A - 5 + 5 = 10 S - 6</p>	<p><math>\cdot \ddot{\text{N}} \cdot \cdot \ddot{\text{N}} \cdot</math></p> <p><math>\cdot \ddot{\text{N}} \cdot \quad \cdot \ddot{\text{N}} \cdot</math></p>	<p><math>:\text{N}:::\text{N}:</math></p> <p><math>:\text{N} \equiv \text{N}:</math></p>
<p>5. C + O (CO<sub>2</sub>)</p> <p>W - 8 + 8 + 8 = 24 A - 4 + 6 + 6 = 16 S - 8</p>	<p><math>\cdot \ddot{\text{O}} \cdot \cdot \ddot{\text{C}} \cdot \cdot \ddot{\text{O}} \cdot</math></p> <p><math>\cdot \ddot{\text{O}} \cdot \quad \cdot \ddot{\text{C}} \cdot \quad \cdot \ddot{\text{O}} \cdot</math></p>	<p><math>\ddot{\text{O}} = \text{C} = \ddot{\text{O}}</math></p>
<p>6. H + O (H<sub>2</sub>O)</p> <p>W - 2 + 2 + 8 = 12 A - 1 + 1 + 6 = 8 S - 4</p>	<p><math>\text{H} \cdot \cdot \ddot{\text{O}} \cdot \cdot \text{H}</math></p> <p><math>\text{H} \cdot \quad \cdot \ddot{\text{O}} \cdot</math></p>	<p><math>\text{H} - \ddot{\text{O}} - \text{H}</math></p>

# Chapter 8 Covalent Bonding Worksheet Answers

## Fruitypiore

**M Lipman**



**Chapter 8 Covalent Bonding Worksheet Answers Fruitypiore:**

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