

## CHEMISTRY REVIEW: KEY CHAPTER 9 – CHEMICAL NAMES AND FORMULAS

- In a Stock name such as iron(II) sulfide, the roman numeral tells us (2)
  - How many atoms of Fe are in one formula unit.
  - How many sulfide ions can be obtained by the ion atom.
  - The charge on each Fe ion.
  - The total positive charge of the formula unit.
- The result of changing a subscript in a correctly written chemical formula is (2)
  - Change the number of ions represented in the formula.
  - Change the charges on the other ions in the compound.
  - Change the formula so that it no longer represents the compound.
  - Have no effect on the formula.
- Derive a generalization for when an acid name will end in the suffix -ic or -ous. If ending in -ic, the anion ended in -ate. If ending in -ous, the anion ended in -ite.
- Derive a generalization for when an acid name will begin with the prefix hydro-. If prefix is hydro-, the anion ended in -ite. This is almost always a monatomic ion.
- Fill in the blanks in the table below.

Compound Name	Formula
Acetic acid	$\text{HC}_2\text{H}_3\text{O}_2$
Aluminum sulfate	$\text{Al}_2(\text{SO}_4)_3$
Lead(II) carbonate	$\text{PbCO}_3$
Ammonium phosphate	$(\text{NH}_4)_3\text{PO}_4$
Sulfuric acid	$\text{H}_2\text{SO}_4$
Nitrogen pentoxide	$\text{N}_2\text{O}_5$
Copper(II) carbonate	$\text{CuCO}_3$
Sodium sulfate	$\text{Na}_2\text{SO}_4$
Ammonium phosphate	$(\text{NH}_4)_3\text{PO}_4$
Mercury sulfate	$\text{HgSO}_4$
Nitrous acid	$\text{HNO}_2$
Magnesium perchlorate	$\text{Mg}(\text{ClO}_4)_2$
Iron(II) chloride	$\text{FeCl}_2$
Iron(III) chloride	$\text{FeCl}_3$
Carbon(IV) oxide	$\text{CO}_2$
Sulfur dioxide	$\text{SO}_2$
Sulfur pentachloride	$\text{SCl}_5$
Tetraphosphorus trisulfide	$\text{P}_4\text{S}_3$
Carbonic acid	$\text{H}_2\text{CO}_3$

- Name the following compound and identify its molecular ions or acid.
  - $\text{Ca}(\text{NO}_3)_2$ , calcium (II) nitrate - II
  - $\text{Fe}(\text{NO}_3)_3$ , Iron(III) nitrate - III
  - $\text{Fe}(\text{OH})_2$ , Iron(II) hydroxide - II
  - $\text{N}_2\text{O}_5$ , Nitrogen pentoxide - III

# Chapter 9 Chemical Names And Formulas Worksheet Answer Key

**Roman Wölfel**



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web may 12 2011 the steam piping must be designed to allow for expansion and contraction of the steam piping the steam lines must have the correct number of appropriately sized hanger supports guides should ensure there are no forces or moments on the tur bine that exceed the values provided by the turbine manufacturer if the piping is unable to flex