

## Student Exploration: Chemical Equations Activity A & B

**Directions:** Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

**Prior Knowledge Questions** (Do these BEFORE using the Gizmo.)

1. A candle is placed on one pan of a balance, and equal weight is placed on the other pan. What would happen if you lit up the candle and waited for a while?

The candle would eventually burn, thus losing mass and becoming lighter. The pan with the other object would lower as the candle burns.

2. Suppose the candle was placed in a large, sealed jar that allowed it to burn for several minutes before running out of oxygen. The candle and jar are balanced by equal weight. In this situation, what would happen if you lit up the candle and waited?

I assume the balance would stay level because the carbon dioxide is trapped within the jar, causing it to retain all of its mass despite the candle burning.

### Gizmo Warm-up

Burning is an example of a **chemical reaction**. The law of **conservation of matter** states that no atoms are created or destroyed in a chemical reaction. Therefore, a balanced **chemical equation** will show the same number of each type of atom on each side of the equation.

To set up an equation in the *Chemical Equations* Gizmo, type the **chemical formulas** into the text boxes of the Gizmo. First, type in "H<sub>2</sub>+O<sub>2</sub>" in the **Reactants** box and "H<sub>2</sub>O" in the **Products** box. This represents the reaction of hydrogen and oxygen gas to form water.



1. Check that the **Visual** display is chosen on each side of the Gizmo, and count the atoms.

A. How many hydrogen atoms are on the **Reactants** side?

2

**Products** side?

2

B. How many oxygen atoms are on the **Reactants** side?

2

**Products** side?

1

2. Based on what you see, is this equation currently balanced, meaning are there **equal** numbers of each reactant and product on both sides of the equation?

This equation is not currently balanced.

### Activity A:

**Interpreting  
chemical  
formulas**

#### Get the Gizmo ready:

- Erase the chemical formulas in each text box.
- Check that the **Visual** displays are selected.



# Chemical Equations Gizmo Teacher Answers

**Jennifer Wilhelm, Ronald  
Wilhelm, Merryn Cole**



## **Chemical Equations Gizmo Teacher Answers:**

**Creating Project-Based STEM Environments** Jennifer Wilhelm, Ronald Wilhelm, Merryn Cole, 2019-02-05 This book models project based environments that are intentionally designed around the United States Common Core State Standards CCSS 2010 for Mathematics the Next Generation Science Standards NGSS Lead States 2013 for Science and the National Educational Technology Standards ISTE 2008 The primary purpose of this book is to reveal how middle school STEM classrooms can be purposefully designed for 21st Century learners and provide evidence regarding how situated learning experiences will result in more advanced learning This Project Based Instruction PBI resource illustrates how to design and implement interdisciplinary project based units based on the REAL Realistic Explorations in Astronomical Learning Unit 1 and CREATES Chemical Reactions Engineered to Address Thermal Energy Situations Unit 2 The content of the book details these two PBI units with authentic student work explanations and research behind each lesson including misconceptions students might hold regarding STEM content pre post research results of unit implementation with over 40 teachers and thousands of students In addition to these two units there are chapters describing how to design one s own research based PBI units incorporating teacher commentaries regarding strategies obstacles overcome and successes as they designed and implemented their PBI units for the first time after learning how to create PBI STEM Environments the REAL way

**Engineering Education** ,1971 *The Ultimate Chemical Equations Handbook, Teacher Edition* Jane D. Smith, George R. Hague, 2011 **Balancing Chemical Equations Worksheet** Crispin Collins, 2020-09-12 Struggling with balancing chemical reaction Balancing chemical equations can look intimidating for lot of us The good news is that practice makes perfect Master balancing skill with this workbook packed with hundreds of practice problems This book is for anyone who wants to master the art of balancing chemical reactions First few chapters of this book are step by step explanation of the concepts and other chapters are for practicing problems This book help students develop fluency in balancing chemical equation which provides plenty of practice Methods to solve with the explanation Total of 550 problems to solve with answer key 450 chemical reactions to practice with answer key 100 practice problems that are needed before balancing a chemical reaction with answer key Click the Buy now button to take advantage of this book to help yourself in mastering balancing skill *Inquiries in Science Chemistry Series* ,2014-10-01 *Inquiries in Science Chemistry Series Balancing Chemical Equations Teacher s Manual Second Edition* **Balancing Chemical Equations Workbook** Kraaya Publishing, 2021-02-21 *Chemical Reactions to Balance Workbook* This chemistry balancing equations practice workbook contains 250 non balanced chemical equations Begin with 2 terms problems Work your way up to 6 terms problems This is the perfect workbook to increase chemistry balancing skills for beginners Table of contents How To Balance A Chemical Equation Chemical Equations To Balance Correct Answers Book features Non repetitive equations Include all reactions types synthesis combustion decomposition Use it now and develop instant recall of balancing equations Enjoy the challenge **Balancing Chemical**

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Handbook, Student Edition Jane D. Smith, George R. Hague, 2011      **Chemistry Equations & Answers** Speedy Publishing, 2014-08-26 General chemistry inorganic chemistry organic chemistry and biochemistry are all difficult courses requiring much memorization for the student Essentially there is no easy way to learn formulas and facts This is why chemistry classes are such challenges to students even the best ones However a chemistry equations and answers study guide can help the student When used as a quick reference guide it can be used often to determine the formulas needed for various questions The astute student can cleverly devise ways to make the guide useful for test questions or other circumstances requiring one of the many chemistry equations      *Balancing Chemical Equations* Learning Hub Publishing, 2020-06-22 To improve their chemistry one must perfect the skill of balancing numerous chemical equations and doing them properly This is something which you will have to practice again and again to perfect as its a key skill when it comes to high school and even in college entrance exam and some college courses In this book you will find multiple practice equations with answers in 3 terms 4 terms 5 terms and many word problems that you can work on and improve your chemistry      **Chemistry Equations and Answers (Speedy Study Guides)** Speedy Publishing LLC, 2015-04-23 Chemistry is a difficult subject to fully comprehend with its equations and scientific laws Trying to digest an entire book in one semester is a tough job but with the help of study guides like these you can absorb information in chemistry much more effectively This guide covers chemical equations including examples potential problems and solutions      *A Balancing Act! How to Balance a Chemical Equation and the Law of Conservation of Mass | Grade 6-8 Physical Science* Baby Professor, 2024-04-15 Explore the intriguing principles of chemical balance with this essential guide perfect for middle school educators homeschooling parents and librarians This book unravels the Law of Conservation of Mass and demystifies the process of balancing chemical equations underscoring their significance in the US STEM curriculum Gain insights into the immutable nature of mass in chemical reactions and master the art of equation balancing to foster a deep understanding of chemical transformations a must have resource for promoting scientific literacy in young learners      **Chemical Equations** James Nixon Pendlebury, 1972      *Writing Chemical Equations and Identifying Unknown Solutions Using Microscale Techniques* Marcia L. Gillette, H. Anthony Neidig, 1992-01-01

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