

Chemistry 530 Midterm Review

Exam on Monday Dec 16 at 8:30

Section I = 75 pts

- Consists of 75 multiple choice questions worth 1 pt. each
- Questions cover all material covered during the second semester
- Questions will be similar to multiple choice problems in the book and review guides on-line
- Practicing the review guides is the best approach; answers will be posted on-line

Section II = 95 pts

- Consists of 19 short answer problems totaling 95 points
- Questions cover the whole semester
- Examples of types of questions
 - Metric Conversions
 - Volume Calculations
 - Percent Composition
 - Atomic Mass
 - Electron configuration
 - Nomenclature
 - Balancing Equations
 - Predicting the products of a reactions
 - Identifying types of reactions
 - Mole Conversions
 - Molar Mass of a compound
 - Molar Mass of a gas at STP
 - Density of a gas at STP
 - Empirical & Molecular Formula

Study approach

- First, read through your notes. Once thoroughly and once more quickly.
- Go through the review packet and make sure you are comfortable with all of the terms and how to do problems
- Do the review packets for help with multiple choice
- Do questions on old tests for practicing written problems
- Make sure you have practiced AT LEAST one of the types of example problems

Chapter 1 – Introduction to Chemistry

- Chemistry: the branch of science that studies the composition and properties of matter.
- Scientific Method:
 - Observation
 - Hypothesis
 - Controlled Experiment
- Theory versus Natural Law

Prerequisite Science Skills

- Rules for Significant Figures:
 - 1) All non-zero digits are significant.
 - 2) Zeros are significant only when they are *between* two non-zero digits or if they occur *after* a decimal point and *after* a non-zero digit.
 - 3) In *multiplication or division*, you cannot end up with more significant figures than you started with.
 - 4) In *addition or subtraction*, you cannot have more digits to the right of the decimal than you started with.
- Exponential numbers
- Scientific Notation = D.DD $\times 10^n$ where $1.00 \leq D.DD \leq 9.99$

Chemistry Midterm Study Guide

**Peter L. Kresan, Reed Mencke, Frank
Press, Raymond Siever**



Chemistry Midterm Study Guide:

Unveiling the Magic of Words: A Overview of "**Chemistry Midterm Study Guide**"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their capability to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "**Chemistry Midterm Study Guide**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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