

6. For the reaction, $A + 2B \rightarrow AB_2$, the order w.r.t. reactant A is 2 and w.r.t. reactant B. What will be change in rate of reaction if the concentration of A is doubled and B is halved?
- increases four times
 - decreases four times
 - increases two times
 - no change
7. Arrange the following in the increasing order of their boiling points:
A : Butanamine, B: N,N-Dimethylethanamine, C: N-Ethylethanamine
- C < B < A
 - A < B < C
 - A < C < B
 - B < C < A
8. The CFSE of $[CoCl_4]^{2-}$ is 18000 cm^{-1} the CFSE for $[CoCl_6]^{4-}$ will be
- 18000 cm^{-1}
 - 9000 cm^{-1}
 - 2000 cm^{-1}
 - 16000 cm^{-1}
9. What would be the major product of the following reaction?
- $$C_6H_5-CH_2-COCl + HBr \rightarrow A + B$$
- A = $C_6H_5CH_2OH$, B = C_6H_5Br
 - A = $C_6H_5CH_2OH$, B = CO_2HBr
 - A = $C_6H_5CH_2Br$, B = C_6H_5OH
 - A = $C_6H_5CH_2Br$, B = C_6H_5OH
10. Which of the following statements is not correct for amines?
- Most alkyl amines are more basic than ammonia solution.
 - pK_a value of ethylamine is lower than benzylamine.
 - CH_3NH_2 (g) reaction with nitrous acid releases NO gas.
 - Hinsberg's reagent reacts with secondary amines to form sulphonamides.
11. Which of the following tests/reactions is given by aldehydes as well as ketones?
- Feigl's test
 - Tollen's test
 - 2,4-DNP test
 - Cannizzaro reaction

Chemistry Multiple Choice Questions And Answers

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of electrolysis reactions conductors and non conductors dry cells electrical devices circuit symbols electrolytes non electrolytes organic solvents polarization and valence electrons The chapter Electricity and Chemicals MCQs covers topics of chemical to electrical energy dry cells electrolyte non electrolyte and polarization The chapter Elements Compounds and Mixtures MCQs covers topics of elements compounds mixtures molecules atoms and symbols for elements **Grade 10 Chemistry Multiple Choice Questions and Answers (MCQs)** Arshad Iqbal,2020-03-13 Grade 10 Chemistry Multiple Choice Questions and Answers MCQs Grade 10 chemistry quizzes practice tests with answer key provides mock tests for competitive exams to solve 842 MCQs Grade 10 Chemistry MCQs helps with theoretical conceptual and analytical study for self assessment career tests This book can help to learn and practice Grade 10 Chemistry quizzes as a quick study guide for placement test preparation Grade 10 Chemistry Multiple Choice Questions and Answers MCQs is a revision guide with a collection of trivia quiz questions and answers on topics Acids bases and salts biochemistry characteristics of acids bases and salts chemical equilibrium chemical industries environmental chemistry atmosphere water hydrocarbons and organic chemistry to enhance teaching and learning Grade 10 Chemistry Quiz Questions and Answers pdf also covers the syllabus of many competitive papers for admission exams of different schools from chemistry textbooks on chapters Acids Bases and Salts Multiple Choice Questions 23 MCQs Biochemistry Multiple Choice Questions 226 MCQs Characteristics of Acids Bases Salts Multiple Choice Questions 115 MCQs Chemical Equilibrium Multiple Choice Questions 46 MCQs Chemical Industries Multiple Choice Questions 67 MCQs Environmental Chemistry I Atmosphere Multiple Choice Questions 97 MCQs Environmental Chemistry II Water Multiple Choice Questions 62 MCQs Hydrocarbons Multiple Choice Questions 87 MCQs Organic Chemistry Multiple Choice Questions 93 MCQs Atmosphere Multiple Choice Questions 26 MCQs The chapter Acids Bases and Salts MCQs covers topics of acids and bases concepts Bronsted concept of acids and bases pH scale and salts The chapter Biochemistry MCQs covers topics of alcohols carbohydrates DNA structure glucose importance of vitamin lipids maltose monosaccharide nucleic acids proteins RNA types of vitamin vitamin and characteristics vitamin and functions vitamin and mineral vitamin deficiency vitamin facts vitamins vitamins and supplements The chapter Characteristics of Acids Bases and Salts MCQs covers topics of concepts of acids and bases pH measurements salts and self ionization of water pH scale The chapter Chemical Equilibrium MCQs covers topics of dynamic equilibrium equilibrium constant and units importance of equilibrium constant law of mass action and derivation of expression and reversible reactions The chapter Chemical Industries MCQs covers topics of basic metallurgical operations metallurgical operations petroleum Solvay process urea and composition The chapter Environmental Chemistry I Atmosphere MCQs covers topics of composition of atmosphere layers of atmosphere stratosphere troposphere ionosphere air pollution environmental issues environmental pollution global warming meteorology and ozone depletion The chapter Environmental Chemistry II Water MCQs covers topics of soft and hard water types of hardness of water water and solvent disadvantages of hard water methods of removing hardness

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