

ANSWER KEY

Chapter 1

Section 1

1. scientia
2. knowledge

Answers 3 and 4 are interchangeable.

3. observation
4. investigation
5. overlap
6. explanations
7. technology
8. modified

Answers 9, 10, and 11 are interchangeable.

9. life science; living things
10. earth science; Earth and space
11. physical science; matter and energy
12. investigations
13. scientific method
14. dependent
15. constant
16. bias

Section 2

- | | |
|-------------|---------------|
| 1. length | 13. kg, mg |
| 2. time | 14. s |
| 3. kilogram | 15. km, m, cm |
| 4. kelvin | 16. a |
| 5. m | 17. a |
| 6. kg | 18. b |
| 7. s | 19. b |
| 8. K | 20. a |

- | | |
|---|--------------------------------|
| 9. m ³ , cm ³ | 21. b |
| 10. mL, cm ³ , L | 22. a |
| 11. g/cm ³ , kg/m ³ | 23. volume = 6 cm ³ |
| 12. K, °C | |

Section 3

1. line graph
2. temperature vs. time for heating of water
3. time
4. horizontal or x -axis
5. vertical or y -axis
6. circle graph
7. the percentage of elements making up living things
8. oxygen
9. bar graph
10. the height of students in Sarah's class
11. 168 cm

Chapter 2

Section 1

1. the total number of m, km, or other distance units that are traveled in total; no; $d = st$
2. the distance from a start point to an end point, with a direction; yes; no formula
3. the total distance traveled in a time interval divided by the time interval; no; $s = d/t$
4. the speed at a certain instant in time; no; no formula
5. the speed with the direction of travel; yes; $s = d/t$ can be used, and a direction added

Chemistry Understanding Main Ideas Part B Answers

Clifford Lane



Chemistry Understanding Main Ideas Part B Answers:

Understanding Basic Chemistry Through Problem Solving: The Learner's Approach (Revised Edition) Kim Seng Chan, Jeanne Tan, 2017-03-02 This book is the revised edition of *Understanding Basic Chemistry Through Problem Solving* published in 2015. It is in a series of *Understanding Chemistry* books which deals with Basic Chemistry using the problem solving approach. Written for students taking either the university of Cambridge O level examinations or the GCSE examinations, this guidebook covers essential topics and concepts under both stipulated chemistry syllabi. The book is written in such a way as to guide the reader through the understanding and applications of essential chemical concepts using the problem solving approach. The authors have also retained the popular discourse feature from their previous few books: *Understanding Advanced Physical Inorganic Chemistry*, *Understanding Advanced Organic and Analytical Chemistry*, *Understanding Advanced Chemistry Through Problem Solving* and *Understanding Basic Chemistry* to help the learners better understand and see for themselves how the concepts should be applied during solving problems. Based on the Socratic Method, questions are implanted throughout the book to help facilitate the reader's development in forming logical conclusions of concepts and the way they are being applied to explain the problems. In addition, the authors have also included important summaries and concept maps to help the learners to recall, remember, reinforce, and apply the fundamental chemical concepts in a simple way.

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[Experiments and Exercises in Basic Chemistry](#) Steven Murov, Brian Stedjee, 2000-02-03 [InfoWorld](#), 1982-02-15

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