

# CHEMISTRY 1

## FINAL EXAM REVIEW

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Name \_\_\_\_\_ Section \_\_\_\_\_ Date \_\_\_\_\_

### Chemistry Review: Writing and Balancing Equations

Directions: Solve each problem and answer each question

1. Balance each equation below by adding the correct coefficients

- $\_\_ \text{CuCl}_2 + \_\_ \text{NaBr} \rightarrow \_\_ \text{NaCl} + \_\_ \text{CuBr}_2$
- $\_\_ \text{HCN} + \_\_ \text{Mg(OH)}_2 \rightarrow \_\_ \text{Mg(CN)}_2 + \_\_ \text{H}_2\text{O}$
- $\_\_ \text{HNO}_3 + \_\_ \text{(NH}_4\text{)}_2\text{SO}_4 \rightarrow \_\_ \text{NH}_4\text{NO}_3 + \_\_ \text{H}_2\text{SO}_4$
- $\_\_ \text{ZnCrO}_4 + \_\_ \text{Al}_2\text{O}_3 \rightarrow \_\_ \text{ZnO} + \_\_ \text{Al}_2\text{CrO}_4$
- $\_\_ \text{C}_2\text{H}_6 + \_\_ \text{O}_2 \rightarrow \_\_ \text{CO}_2 + \_\_ \text{H}_2\text{O}$

2. Write and balance each equation. Include phase labels

- Aqueous sodium hydroxide reacts with aqueous sulfuric acid to form aqueous sodium sulfate and water
- Solid iron (III) hydroxide and aqueous sodium carbonate are formed from the reaction of solid iron (III) carbonate and aqueous sodium hydroxide
- Solid aluminum reacts with aqueous hydrobromic acid to form aqueous aluminum bromide and hydrogen gas
- Gaseous ethane (dibarbon hexahydride) reacts with oxygen gas to form gaseous carbon dioxide and gaseous water
- Solid calcium phosphite reacts with aqueous copper(II) nitrate to form aqueous calcium nitrate and solid copper (II) phosphite

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