

# NUMBER OF ATOMS IN A FORMULA

Name \_\_\_\_\_

Determine the number of atoms in the following chemical formulas.

1. NaCl \_\_\_\_\_

2. H<sub>2</sub>SO<sub>4</sub> \_\_\_\_\_

3. KNO<sub>3</sub> \_\_\_\_\_

4. CaCl<sub>2</sub> \_\_\_\_\_

5. C<sub>2</sub>H<sub>6</sub> \_\_\_\_\_

6. Ba(OH)<sub>2</sub> \_\_\_\_\_

7. NH<sub>4</sub>Br \_\_\_\_\_

8. Ca<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub> \_\_\_\_\_

9. Al<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub> \_\_\_\_\_

10. Mg(NO<sub>3</sub>)<sub>2</sub> \_\_\_\_\_

11. Cu(NO<sub>3</sub>)<sub>2</sub> \_\_\_\_\_

12. KMnO<sub>4</sub> \_\_\_\_\_

13. H<sub>2</sub>O<sub>2</sub> \_\_\_\_\_

14. H<sub>3</sub>PO<sub>4</sub> \_\_\_\_\_

15. (NH<sub>4</sub>)<sub>3</sub>PO<sub>4</sub> \_\_\_\_\_

16. Fe<sub>2</sub>O<sub>3</sub> \_\_\_\_\_

17. NaC<sub>2</sub>H<sub>3</sub>O<sub>2</sub> \_\_\_\_\_

18. Mg(C<sub>2</sub>H<sub>3</sub>O<sub>2</sub>)<sub>2</sub> \_\_\_\_\_

19. Hg<sub>2</sub>Cl<sub>2</sub> \_\_\_\_\_

20. K<sub>2</sub>SO<sub>3</sub> \_\_\_\_\_

# Dot Point Chemistry Answers Bonding

**Lei Shi**

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