

Guided Practice - Acids & Bases**Properties of Acids and Bases**

1. Match the following terms with their descriptions:

- | | | |
|------------------------|-------|---|
| a. Arrhenius Acid | _____ | the substance formed when a proton is added to _____ |
| b. Arrhenius Base | _____ | the remaining substance when a proton is lost _____ |
| c. Conjugate Acid | _____ | from an acid _____ |
| d. Conjugate Base | _____ | a substance that produces OH ⁻ in solution _____ |
| e. Bronsted-Lowry Acid | _____ | a substance producing H ⁺ in solution _____ |
| f. Bronsted-Lowry Base | _____ | a proton acceptor _____ |
| | | a proton donor _____ |

2. Which of the following represent conjugate acid-base pairs? Label each species as a conjugate acid or a conjugate base. For those pairs that are NOT conjugates, write the correct conjugate acid or base for each species in the pair, labeling each species correctly.

- HSO_4^- ; SO_4^{2-}
- HBr ; BrO^-
- H_3PO_4 ; PO_4^{3-}
- H_2O ; HO^-

3. Complete the following table regarding acid strength.

Property	Strong Acid	Weak Acid
Definition		
Describe Dissociation		
Species present in solution		
Relative strength of conjugate base compared to water		
Equation showing acid's reaction to water		

Chapter 20 Acids And Bases Answer Key

SA Adler



Chapter 20 Acids And Bases Answer Key:

Chemistry (Teacher Guide) Dr. Dennis Englin, 2018-02-26 This book was created to help teachers as they instruct students through the Master's Class Chemistry course by Master Books The teacher is one who guides students through the subject matter helps each student stay on schedule and be organized and is their source of accountability along the way With that in mind this guide provides additional help through the laboratory exercises as well as lessons quizzes and examinations that are provided along with the answers The lessons in this study emphasize working through procedures and problem solving by learning patterns The vocabulary is kept at the essential level Practice exercises are given with their answers so that the patterns can be used in problem solving These lessons and laboratory exercises are the result of over 30 years of teaching home school high school students and then working with them as they proceed through college Guided labs are provided to enhance instruction of weekly lessons There are many principles and truths given to us in Scripture by the God that created the universe and all of the laws by which it functions It is important to see the hand of God and His principles and wisdom as it plays out in chemistry This course integrates what God has told us in the context of this study Features Each suggested weekly schedule has five easy to manage lessons that combine reading and worksheets Worksheets quizzes and tests are perforated and three hole punched materials are easy to tear out hand out grade and store Adjust the schedule and materials needed to best work within your educational program Space is given for assignments dates There is flexibility in scheduling Adapt the days to your school schedule Workflow Students will read the pages in their book and then complete each section of the teacher guide They should be encouraged to complete as many of the activities and projects as possible as well Tests are given at regular intervals with space to record each grade About the Author DR DENNIS ENGLIN earned his bachelor's from Westmont College his master of science from California State University and his EdD from the University of Southern California He enjoys teaching animal biology vertebrate biology wildlife biology organismic biology and astronomy at The Master's University His professional memberships include the Creation Research Society the American Fisheries Association Southern California Academy of Sciences Yellowstone Association and Au Sable Institute of Environmental Studies

Physical Chemistry for Engineering and Applied Sciences Frank R. Foulkes, 2012-09-12 Physical Chemistry for Engineering and Applied Sciences is the product of over 30 years of teaching first year Physical Chemistry as part of the Faculty of Applied Science and Engineering at the University of Toronto Designed to be as rigorous as compatible with a first year student's ability to understand the text presents detailed step by step derivations of the equations that permit the student to follow the underlying logic and of equal importance to appreciate any simplifying assumptions made or mathematical tricks employed In addition to the 600 exercises and end of chapter problems the text is rich in worked non trivial examples many of which are designed to be inspiring and thought provoking Step by step derivation of all equations enables the student to smoothly follow the derivation by sight and can be understood relatively easily by students with

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Textbook of Biochemistry for Dental Students
DM Vasudevan, Sreekumari S, Kannan Vaidyanathan, 2017-12-31 The third edition of this textbook has been fully revised to provide dental students with the latest information in the field of biochemistry The comprehensive text is divided into 31 chapters and covers key aspects of biochemistry that students will need to know for examinations including amino acids carbohydrates lipids metabolism acid base balance dental caries and much more The new edition features a question bank of essay type and short note questions based on previous examination papers The book is highly illustrated with clinical images tables and boxes and key points in each chapter are highlighted for quick reference and to assist learning Key points Fully revised third edition providing dental students with latest information in the field of biochemistry Features question bank of essay type and short note practice questions Highlights key learning points for each topic Previous edition 9789350254882 published in 2011

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Chemistry Theodore Lawrence Brown, 1988

Journal of the Society of Chemical Industry Society of Chemical Industry (Great Britain), 1883

Watts' Dictionary of Chemistry, Revised and Entirely Rewritten Henry Watts, 1905

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