

## Chapter Test B

### Chapter: Stoichiometry

**PART I** In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- \_\_\_\_\_ 1. Knowing the mole ratio of a reactant and product in a chemical reaction would allow you to determine
- the energy released in the reaction.
  - the speed of the reaction.
  - the mass of the product produced from a known mass of reactant.
  - whether the reaction is reversible.
- \_\_\_\_\_ 2. In the equation  $6\text{CO}_2 + 6\text{H}_2\text{O} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2$ , the mole ratio of water to oxygen is
- 2:1.
  - 1:1.
  - 3:2.
  - 6:8.
- \_\_\_\_\_ 3. In the equation  $\text{Al}_2(\text{SO}_4)_3 + 3\text{Ca}(\text{OH})_2 \rightarrow 3\text{CaSO}_4 + 2\text{Al}(\text{OH})_3$ , the mole ratio of calcium hydroxide to aluminum hydroxide is
- 1:3.
  - 2:3.
  - 1:1.
  - 3:2.
- \_\_\_\_\_ 4. For the reaction represented by the equation  $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$ , how many moles of nitrogen are required to produce 18 mol of ammonia?
- 9
  - 18
  - 27
  - 36
- \_\_\_\_\_ 5. In the reaction represented by the equation  $\text{A} + \text{B} \rightarrow \text{C} + \text{D}$ , if there is more reactant B than is required to completely react with all of A, then
- A is the limiting reactant.
  - B is the limiting reactant.
  - there is no limiting reactant.
  - no product can be formed.

# Chapter 9 Stoichiometry Test Answers

**Yi-Tong Ma**



## Chapter 9 Stoichiometry Test Answers:

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**SAT II** Claire Aldridge, Karla Lee, Kaplan, Inc, 2000 Expert review prepares test takers for all the chemistry content they need to know with three full length practice tests and detailed explanations Charts

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antisemitism threatens the security of jewish people and communities around the world and is the driving force of many

violent extremist ideologies like all forms of intolerance and discrimination antisemitism has a profound impact on the whole of society

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