

Thermodynamic Equilibrium - 1 (Multiple choice)

Multiple choice

Identify the correct statement about each condition for thermodynamic equilibrium between two systems.

1. The temperatures of two systems are different, but the (thermodynamic) entropies of the systems are equal.
- True
- False
- All of the above.
2. The two thermodynamic states have the same pressure and temperature, but not the same volume.
- True
- False
3. The two systems are both expanding, but the left system expands more rapidly than the right system.
- True
- False
- None of the above.
4. The two systems are both expanding, but the left system expands more slowly than the right system.
- True
- False
- None of the above.
5. The two thermodynamic states are at the same pressure, temperature, volume and chemical composition.
- Equilibrium is reached at once.
- Equilibrium is not reached.
- Equilibrium is reached after a long time.
- Equilibrium is never reached.
6. The two systems are both expanding, but the left system expands more slowly.
- True
- False
- None of the above.
7. The two systems are both expanding, but the left system expands more rapidly.
- True
- False
- None of the above.
8. The two systems are both expanding, but the left system expands more slowly, and the right system is contracting.
- Equilibrium is reached at once.
- Equilibrium is not reached.
- Equilibrium is reached after a long time.
- Equilibrium is never reached.

Chapter Test Thermal Energy Answers

SB Merriam

A decorative graphic element consisting of a light blue horizontal bar with a rounded right end, and a red-to-white gradient semi-circle on the right side of the bar.

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