

**BISHOP HARTMANN ACADEMY +2**  
**FIRST TERMINAL EXAMINATION**

(2020-21)

**SCIENCE (PAPER -II)**  
**CHEMISTRY (THEORY-80 MARKS)**  
**CLASS-IX**  
**(TWO HOURS)**

Answer to this Paper must be written on the paper provided separately.

You will ~~not~~ be allowed to write during the first 15 minutes.

This time is to be spent in reading the Question Paper.

The time given at the head of this paper is the time allowed for writing the answers.

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**Section - I** is compulsory. Attempt **any four** questions from **Section- II**.

The intended marks for questions or parts of questions are given in brackets [ ].

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**SECTION- I ( 40 MARKS)**

Attempt **all** questions from this section

**Question-1**

**A.** Fill in the blanks from the choices given in brackets :

[5]

- (i) Sodium chloride has \_\_\_\_\_ (2/3/4) radicals.
- (ii) Formula of iron (III) carbonate is \_\_\_\_\_ [  $FeCO_3$  /  $Fe_2CO_3$  /  $Fe_2(CO_3)_3$  ].
- (iii) In the type of reaction called \_\_\_\_\_ (displacement/decomposition/double decomposition) two compounds exchange their positive and negative radicals.
- (iv)  $HgSO_4 \cdot 7H_2O$  is commonly known as \_\_\_\_\_ (Epsom salt/Glauber's salt/gypsum).
- (v) Atoms having same mass number but different atomic numbers are known as \_\_\_\_\_ (isotopes/isobars/isotones).

**B.** Choose the correct answer from the options given below:

[2]

- (i) Number of valence electrons in  $O^{2-}$  is :
  - (a) 6
  - (b) 8
  - (c) 10
  - (d) 4

# Chemistry Hl Paper 2

**Eugene Franz Roeber, Howard Coon  
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