

Chemical Analysis – What can you Remember?

1. What colour precipitate do chloride ions produce?
 - a) White
 - b) Cream
 - c) Yellow
 - d) Green
2. What colour do sodium ions go in a flame test?
 - a) Red
 - b) Yellow
 - c) Brick-red
 - d) Lilac
3. Which two solutions are added to test for the presence of halide ions?
 - a) Silver chloride and nitric acid
 - b) Barium chloride and hydrochloric acid
 - c) Silver nitrate and nitric acid
 - d) Barium chloride and sulphuric acid
4. What is the test for oxygen gas?
 - a) Lit splint goes out with a squeaky pop
 - b) Damp blue litmus paper goes red
 - c) Damp red litmus paper goes blue
 - d) Glowing splint relights
5. What colour precipitate do iodide ions form?
 - a) White
 - b) Cream
 - c) Yellow
 - d) Green
6. Which ion produces a brick-red colour in a flame test?
 - a) Sodium
 - b) Calcium
 - c) Copper
 - d) Lithium
7. What is the test for carbon dioxide gas?
 - a) Limewater turns cloudy
 - b) Damp blue litmus turns red
 - c) A glowing splint relights
 - d) A lit splint goes out with a squeaky pop
8. When sodium hydroxide is added to a solution containing iron (III) ions, what colour precipitate forms?
 - a) White
 - b) Blue
 - c) Green
 - d) Red/brown
9. What is the test for chlorine gas?
 - a) Damp blue litmus paper turns red and is then bleached white
 - b) Red litmus \square turns blue
 - c) A glowing splint relights
 - d) Limewater will turn cloudy
10. Copper (II) ions form what colour precipitate test?
 - a) Blue
 - b) Green
 - c) Red/brown
 - d) Cream
11. Which two solutions are added to test for the presence of sulphate ions?
 - a) Silver chloride and nitric acid
 - b) Barium chloride and sulphuric acid
 - c) Silver nitrate and nitric acid
 - d) Barium chloride and hydrochloric acid
12. What colour do potassium ions go in a flame test?
 - a) Lilac
 - b) Red
 - c) Yellow
 - d) Brick-red
13. What is the test for hydrogen gas?
 - a) Lit splint goes out with a squeaky pop
 - b) Glowing splint relights
 - c) Damp blue litmus paper turns red and then is bleached white
 - d) Limewater goes cloudy
14. Which ion produces a red colour in a flame test?
 - a) Sodium
 - b) Calcium
 - c) Copper
 - d) Lithium
15. What colour precipitate do bromide ions form?
 - a) White
 - b) Cream
 - c) Yellow
 - d) Green
16. What needs to be added to test for the presence of carbonate ions?
 - a) Silver nitrate
 - b) Universal indicator
 - c) Litmus paper
 - d) Dilute acid
17. If sodium hydroxide is added to a compound containing ammonium ions, what gas is produced?
 - a) Oxygen
 - b) Ammonia
 - c) Carbon dioxide
 - d) Hydrogen
18. What colour precipitate does barium sulphate form?
 - a) White
 - b) Green
 - c) Yellow
 - d) Cream
19. What is needed to carry out a flame test?
 - a) Splint
 - b) Glass rod
 - c) Spatula
 - d) Nichrome/platinum wire

Your score out of 19 =
Percentage =

Chemistry Unit 2 Test Answers

Charles Watters Odell



Chemistry Unit 2 Test Answers:

The book delves into Chemistry Unit 2 Test Answers. Chemistry Unit 2 Test Answers is a crucial topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Chemistry Unit 2 Test Answers, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Chemistry Unit 2 Test Answers
 - Chapter 2: Essential Elements of Chemistry Unit 2 Test Answers
 - Chapter 3: Chemistry Unit 2 Test Answers in Everyday Life
 - Chapter 4: Chemistry Unit 2 Test Answers in Specific Contexts
 - Chapter 5: Conclusion
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5. In chapter 4, this book will scrutinize the relevance of Chemistry Unit 2 Test Answers in specific contexts. This chapter will explore how Chemistry Unit 2 Test Answers is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Chemistry Unit 2 Test Answers. This chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Chemistry Unit 2 Test Answers.

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Table of Contents Chemistry Unit 2 Test Answers

1. Understanding the eBook Chemistry Unit 2 Test Answers
 - The Rise of Digital Reading Chemistry Unit 2 Test Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Unit 2 Test Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an eBook Platform
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Unit 2 Test Answers
 - Personalized Recommendations
 - Chemistry Unit 2 Test Answers User Reviews and Ratings
 - Chemistry Unit 2 Test Answers and Bestseller Lists
5. Accessing Chemistry Unit 2 Test Answers Free and Paid eBooks
 - Chemistry Unit 2 Test Answers Public Domain eBooks
 - Chemistry Unit 2 Test Answers eBook Subscription Services
 - Chemistry Unit 2 Test Answers Budget-Friendly Options
6. Navigating Chemistry Unit 2 Test Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry Unit 2 Test Answers Compatibility with Devices
 - Chemistry Unit 2 Test Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Unit 2 Test Answers
 - Highlighting and Note-Taking Chemistry Unit 2 Test Answers
 - Interactive Elements Chemistry Unit 2 Test Answers
8. Staying Engaged with Chemistry Unit 2 Test Answers

- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Unit 2 Test Answers
9. Balancing eBooks and Physical Books Chemistry Unit 2 Test Answers
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Unit 2 Test Answers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Unit 2 Test Answers
- Setting Reading Goals Chemistry Unit 2 Test Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Unit 2 Test Answers
- Fact-Checking eBook Content of Chemistry Unit 2 Test Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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