

EXERCISE SEQUENCE

Building on the exercises in the preceding chapter, the intermediate Programme adds several new exercises to your workout. This exercise sequence chart provides a visual cue card: when performed in the order shown, the entire programme will flow smoothly and easily. As you progress, you should need to refer only to the chart to complete your workout.



1 The Hundred
0:45-1:00

2 Roll Up
0:30-1:15

3 Single Leg Circles
0:30-1:00



4 Rolling Like a Ball
0:45-1:00

5 Single Leg Stretch
0:30-1:15

6 Double Leg Stretch
0:30-1:00



7 Single Straight Leg Stretch
0:30-1:00

8 Double Straight Leg Stretch
0:30-1:00

9 One-Down
0:30-1:00



10 Spine Stretch Forward
0:30-1:00



11 Open Leg Rocker
0:30-1:00



12 Corkscrew
0:45-1:00



13 Saw
0:45-1:00



14 Neck Roll
0:30-1:00



15 Single Leg Kick
0:30-1:00



16 Double Leg Kick
0:30-1:00



17 Neck Pull
0:30-1:00



18 Side Kicks Series
0:30-1:00



19 Tractor Series
0:30-1:00



20 Seal
0:30-1:00

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web answer the following questions based on the solubility curve below which salt is least soluble in water at 20 c 2 how

many grams of potassium chloride can be dissolved in 200 g of water at 80 c io 3 at 40 c how much potassium nitrate coin be dissoiu tl n 30d g of water w 4 which salt shows the least change

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web different substances q why do temperatures range from 0 100 c a water is a liquid at these temperatures solubility curves tell us how much solid will dissolve at a given temperature they also allow certain calculations

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web chemistry solubility curves virtual lab curves to clearly show the solubility of the four ionic compounds in the data table below the following tips will help you make the best possible graph make the graph large it should fill three fourths of the graph paper

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web solvents the larger volume of the mixture and solutes the smaller volume of the mixture for example a hot chocolate is a solution in which the solute the chocolate powder is dissolved in the solvent the milk or water the solute and solvent can be

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web question solubility curve of potassium nitrate purpose in this experiment you will determine the solubilities of potassium nitrate in water at various temperatures and construct a solubility curve for knoy

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web they should be able to come up with the following answers stir the water increase the temperature and increase the amount of water alternately you may challenge the students to a contest

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web use a solubility curve to determine whether a solution is saturated or unsaturated and to estimate the amount of excess solute that remains undissolved in the solution

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web give results of all calculations in blue cells you must put your answers in the indicated cells as the grade book pulls your data from those cells do not put any units in the number cells use the unit boxes given graph tab transfer from data page the temp and concentration of salt in units of g salt 100 g water

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of a solid purpose to determine how the solubility of a salt ammonium chloride varies with temperature in order to create a solubility curve based on observed data

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web 30 09 99 solubility curves class data write down the data answer questions on bottom of the sheet results of lab solubility of KClO_3 unsaturated saturation and solubility curves in this experiment we expressed concentration in units of g solute g H_2O putting concentrations in g solute g solution would be tedious and impractical see fig 2

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web use the solubility curve to answer these questions determine how much potassium nitrate will dissolve in 20 g of water at 40 c 200 cm³ of saturated lead ii nitrate solution was prepared at a temperature of 90 c

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web collect experimental data and create a solubility curve by the end of this lab students should be able to properly use an analytical balance to measure mass set up an experimental work station to measure the solubility of a salt in water as a function of the temperature generate a workbook using google sheets prior knowledge

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web sample answer sugar has a much greater solubility in hot water than cold water so its solubility curve probably resembles the sloping solubility curve of potassium nitrate temperature solubility g 100 ml 10 c 36 g 100 ml 20 c 36 g 100 ml 30 c 37 g 100 ml 40 c 37 g 100 ml 50 c 38 g 100 ml 60 c 38 g 100 ml 70 c 39 g 100 ml 80

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web may 29 2018 figure 2 the graph shows the solubility curve of potassium nitrate with corresponding temperatures as the temperature increases so does the solubility of the solution a y 0 2 1 61 b y 0 2 1 61 0 60 2 1 60 61 0 40 2 1 40 61 8 5 g 100 ml 4 6 g 100 ml

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web mar 15 2021 this experiment will further develop your understanding of thermodynamics while relating it to two concepts you already have studied equilibrium and solubility you will determine three thermodynamic values associated with the solubility of potassium nitrate the reaction we will be studying is the dissolution of potassium nitrate in water

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web analysis construct a solubility curve using the mass of solute that will dissolve in 100 g of water at a specified temperature in this lab 5 ml of water was used assume 1.0 g/ml of water take the collected data make it proportional to get the solubility of the KNO_3 graph temperature vs solubility using a best fit curve

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