

Name: .....

### ATOMIC STRUCTURE

1 Use the words in the box to answer the following questions.

electron, neutron, nucleus, proton, shells



- (a) The central part of an atom is called the .....
- (b) The ..... has a negative charge
- (c) The ..... has no charge
- (d) The nucleus contains the ..... and the .....
- (e) Electrons are found in .....
- (f) The ..... and the ..... have the same mass.

2 An atom contains 32 protons and 48 neutrons.

- (a) What is its atomic number? .....
- (b) How many electrons does it have? .....
- (c) What is the mass of the atom? .....

3 Complete the table to show the missing numbers of particles in the atoms A - E.

Atom	Protons	Neutrons	Electrons	Mass
A	20			40
B			21	45
C		24		44
D	19	21		
E			22	45

Which two atoms are isotopes of the same element? ..... and .....

4 Complete the table to show the missing information:

Atom	Atomic number	Electron arrangement	Group in the Periodic Table	Period in the Periodic Table
P		2, 8, 7		
Q	11			
R			4	2
S			3	3

# Atomic Structure Test Questions

**SJ Ball**



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## **Table of Contents Atomic Structure Test Questions**

1. Understanding the eBook Atomic Structure Test Questions
  - The Rise of Digital Reading Atomic Structure Test Questions
  - Advantages of eBooks Over Traditional Books
2. Identifying Atomic Structure Test Questions
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Atomic Structure Test Questions
  - User-Friendly Interface
4. Exploring eBook Recommendations from Atomic Structure Test Questions
  - Personalized Recommendations
  - Atomic Structure Test Questions User Reviews and Ratings
  - Atomic Structure Test Questions and Bestseller Lists
5. Accessing Atomic Structure Test Questions Free and Paid eBooks
  - Atomic Structure Test Questions Public Domain eBooks
  - Atomic Structure Test Questions eBook Subscription Services
  - Atomic Structure Test Questions Budget-Friendly Options
6. Navigating Atomic Structure Test Questions eBook Formats

- ePub, PDF, MOBI, and More
- Atomic Structure Test Questions Compatibility with Devices
- Atomic Structure Test Questions Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Atomic Structure Test Questions
  - Highlighting and Note-Taking Atomic Structure Test Questions
  - Interactive Elements Atomic Structure Test Questions
- 8. Staying Engaged with Atomic Structure Test Questions
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Atomic Structure Test Questions
- 9. Balancing eBooks and Physical Books Atomic Structure Test Questions
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Atomic Structure Test Questions
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Atomic Structure Test Questions
  - Setting Reading Goals Atomic Structure Test Questions
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Atomic Structure Test Questions
  - Fact-Checking eBook Content of Atomic Structure Test Questions
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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