



7. Which of the following is the atomic number of the atom represented by the diagram on the left?

- A. 22
- B. 25
- C. 47
- D. 5

8. What is the total electrical charge for the nucleus of an atom with 8 protons, 2 neutrons, and 10 electrons?

- A. +12
- B. -10
- C. +8
- D. -2

9. Which of the three following atoms has the greatest atomic mass?



1



2



3

- A. atom 1
- B. atom 2
- C. atom 3
- D. They all have equal atomic mass.

10. Which correctly describes the charges of the subatomic particles of an atom?

- A. electrons/negative, protons/neutral, neutrons/positive
- B. protons/positive, neutrons/neutral, electrons/negative
- C. neutrons/negative, neutrons/neutral, electrons/positive
- D. neutrons/negative, protons/positive, electrons/neutral

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