

Dilutions – Practice Problems

Instructions: Use the equation for dilutions to complete the problems below.

- 1) A student obtains 0.30 liters of 2.0 M silver nitrate solution. She dilutes the solution until it has a volume of 2.6 liters. What is the **molarity** of the new solution?

$$M_1 V_1 = M_2 V_2$$

$$(0.6) = M_2 (2.6)$$

$$(2.0) (0.30) = M_2 (2.6)$$

$$M_2 = 0.23 \text{ M}$$

- 2) Before an experiment, a scientist obtains 2.45 L of 1.25 M sulfuric acid (H₂SO₄) and dilutes it until the molarity drops to 0.80 M. What is the **volume** of the new solution?

$$M_1 V_1 = M_2 V_2$$

$$(3.06) = (0.8) V_2$$

$$(1.25) (2.45) = (0.8) V_2$$

$$V_2 = 3.83 \text{ L}$$

- 3) A solution of hydrochloric acid is diluted, bringing the volume from 2.0 L to 8.5 L. If the solution has a final molarity of 1.4 M, what was the **molarity before** dilution?

$$M_1 V_1 = M_2 V_2$$

$$M_1 (2.0) = 11.9$$

$$M_1 (2.0) = (1.4) (8.5)$$

$$M_1 = 5.95 \text{ M}$$

- 4) If a student adds 250 mL of water to a 300 mL solution of NaCl that has a concentration of 3.0 M, what will be the **molarity** of the final solution?

$$M_1 V_1 = M_2 V_2$$

$$(900) = M_2 (550)$$

$$(3.0) (300) = M_2 (550)$$

$$M_2 = 1.64 \text{ M}$$



- 5) A scientist adds 1.2 liters of water to a 5.0 liter solution, causing the molarity to become 0.4 M. What was the **molarity of the initial solution** before the dilution?

$$M_1 V_1 = M_2 V_2$$

$$M_1 (5.0) = 2.48$$

$$M_1 (5.0) = (0.4) (6.2)$$

$$M_1 = 0.50 \text{ M}$$

Solutions Molarity And Dilution Practice Answer Key

Mark Steyvers



Solutions Molarity And Dilution Practice Answer Key:

This book delves into Solutions Molarity And Dilution Practice Answer Key. Solutions Molarity And Dilution Practice Answer Key is a crucial topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Solutions Molarity And Dilution Practice Answer Key, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Solutions Molarity And Dilution Practice Answer Key
- Chapter 2: Essential Elements of Solutions Molarity And Dilution Practice Answer Key
- Chapter 3: Solutions Molarity And Dilution Practice Answer Key in Everyday Life
- Chapter 4: Solutions Molarity And Dilution Practice Answer Key in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Solutions Molarity And Dilution Practice Answer Key. This chapter will explore what Solutions Molarity And Dilution Practice Answer Key is, why Solutions Molarity And Dilution Practice Answer Key is vital, and how to effectively learn about Solutions Molarity And Dilution Practice Answer Key.

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4. In chapter 3, the author will examine the practical applications of Solutions Molarity And Dilution Practice Answer Key in daily life. This chapter will showcase real-world examples of how Solutions Molarity And Dilution Practice Answer Key can be effectively utilized in everyday scenarios.

5. In chapter 4, this book will scrutinize the relevance of Solutions Molarity And Dilution Practice Answer Key in specific contexts. The fourth chapter will explore how Solutions Molarity And Dilution Practice Answer Key is applied in specialized fields, such as education, business, and technology.

6. In chapter 5, the author will draw a conclusion about Solutions Molarity And Dilution Practice Answer Key. The final chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Solutions Molarity And Dilution Practice Answer Key.

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