

Chapter 4 – Atomic Structure

Short Answer: Answer the following questions in complete sentences using science vocabulary. Make sure that your answers are complete and detailed.

1. A sample of calcium contains calcium-40, calcium-44, calcium-42, calcium-48, calcium-43, and calcium-46 atoms. Explain why these atoms have different mass numbers but must have the same atomic number.

The atoms have different mass numbers because they have a different number of neutrons. The mass number is equaled to the number of protons plus neutrons.

The atoms must have the same atomic number because the atomic number is equaled to the number of protons. The number of protons is unique for each element.

2. Complete the following table. Circle the name of the element(s) that are isotopes. Complete Bohr's Model diagrams for the element with eleven electrons and the element with an atomic number of 9.

Element Name	Element Symbol	Atomic Number	Mass Number	Protons	Neutrons	Electrons
Scandium	Sc	21	45	21	24	21
Fluorine	F	9	19	9	10	9
Sodium	Na	11	24	11	13	11
Answers Vary					82	

Chapter 4 Atomic Structure Worksheet Answers

RS Peters



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Earth Science Samuel N. Namowitz, Nancy E. Spaulding, 1989 *Building Block* Prentice-Hall Staff, 1994 Exercises for Principles of Imaging Science and Protection Michael A. Thompson, Steven B. Dowd, Janice D. Hall, Marian P. Hattaway, 1994 A smart collection of additional review questions and practise problems to further reinforce basic concepts The laboratory manual section includes about 30 basic experiments which concentrate more on radiographic factors which affect the quality of the resulting radiograph Readers with a weaker math science background will appreciate the remediation sections and self tests conclude each section to gauge the reader's weaknesses *Simplified ICSE Chemistry* Viraf J. Dalal, **Discover Science: Teacher's annotated edition**, 1991 Science content helps develop the skills needed to understand how science works learn new concepts solve problems and make decisions in today's technological society **Merrill Chemistry** Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 *Special Events* Alan L. Wendroff, 1999-01-28 Providing a concrete outline on event planning within the larger framework for achieving the organizational goals this book introduces the building blocks for a successful special event or any fund raising programme including raising money marketing and public relations building and expanding the non profit donor base building and motivating the volunteer pool training and motivating your board to solicit funds and educating the public about your organization and its mission **Te HS&T J** Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004-02 Te HS&T 2007 Shrt Crs M Holt Rinehart & Winston, 2007 *Glencoe Science*, 2002 Holt Science and Technology Holt Rinehart & Winston, 2004-02 **Science Insights**, 1999 **Exploring Earth and Space** Michael DiSpezio, 1995 A textbook exploring such aspects of matter and energy as heat electricity and nuclear chemistry with suggested activities and review questions at the end of each chapter *Addison-Wesley Science Insights*, 1996

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