



## Reinforcement

# Stability in Bonding

**Directions:** Each statement below contains a pair of terms or phrases in parentheses. Circle the term or phrase that makes each statement true.

- The properties of a compound are (the same as, different from) the properties of the elements that make up the compound.
- Na and Cl are chemical (symbols, formulas).
- NaCl and NaOH are chemical (symbols, formulas).
- In the formula  $H_2O$ , the number 2 is a (subscript, superscript).
- The number 2 in the formula  $H_2O$  tells you that each unit of this compound contains two (hydrogen, oxygen) atoms.
- If a symbol in a chemical formula does not have a subscript after it, a unit of that compound contains (no atoms, one atom) of that element.
- The total number of atoms in  $Fe_2O_3$  is (two, five, six).
- There are (three, seven, ten) different elements in  $H_2SO_4$ .
- An atom is chemically stable if its outer energy level (is filled with, contains no) electrons.
- For atoms of most noble gases and most other elements, the outer energy level is full when it has (3, 8) electrons.
- The noble gases do not readily form compounds because they (are, are not) chemically stable.
- A chemical bond is a (force, chemical) that holds atoms together in a compound.
- Chemical bonds form when atoms lose, gain, or (share, multiply) electrons.

**Directions:** Complete the table below by using the formula of each compound to identify the elements that each compound contains and the number of atoms of each of these elements in a unit of the compound. The first formula has been done for you.

	Formula	Element 1	Element 2	Element 3
	$H_2O$	2 hydrogen	1 oxygen	
14.	NaOH			
15.	NaCl			
16.	$NH_3$			
17.	$H_2SO_4$			
18.	$SiO_2$			

# Section 1 Reinforcement Stability In Bonding Answers

**G Psacharopoulos**

A decorative graphic element consisting of a light blue horizontal bar with a rounded right end, and a red-to-white gradient semi-circle overlapping the right side of the bar.

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James E. Ambrose, 1989

## **Section 1 Reinforcement Stability In Bonding Answers** Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Section 1 Reinforcement Stability In Bonding Answers**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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