

Stoichiometry

Chem 1019 Section 1

Section 10: Introduction to Stoichiometry

In the space provided, write the letter of the best response that best completes each problem or best answers each question.

1. The number of moles of H_2 that can be produced from 100.0 g of H_2O is:
- a. 5.56 mol
 - b. 5.56 mol
 - c. 11.12 mol
 - d. 22.24 mol
2. The number of moles of H_2 that can be produced from 100.0 g of H_2O is:
- a. 5.56 mol
 - b. 11.12 mol
 - c. 22.24 mol
 - d. 44.48 mol
3. The number of moles of H_2 that can be produced from 100.0 g of H_2O is:
- a. 5.56 mol
 - b. 11.12 mol
 - c. 22.24 mol
 - d. 44.48 mol
4. The number of moles of H_2 that can be produced from 100.0 g of H_2O is:
- a. 5.56 mol
 - b. 11.12 mol
 - c. 22.24 mol
 - d. 44.48 mol
5. The number of moles of H_2 that can be produced from 100.0 g of H_2O is:
- a. 5.56 mol
 - b. 11.12 mol
 - c. 22.24 mol
 - d. 44.48 mol
6. The number of moles of H_2 that can be produced from 100.0 g of H_2O is:
- a. 5.56 mol
 - b. 11.12 mol
 - c. 22.24 mol
 - d. 44.48 mol
7. The number of moles of H_2 that can be produced from 100.0 g of H_2O is:
- a. 5.56 mol
 - b. 11.12 mol
 - c. 22.24 mol
 - d. 44.48 mol
8. The number of moles of H_2 that can be produced from 100.0 g of H_2O is:
- a. 5.56 mol
 - b. 11.12 mol
 - c. 22.24 mol
 - d. 44.48 mol
9. The number of moles of H_2 that can be produced from 100.0 g of H_2O is:
- a. 5.56 mol
 - b. 11.12 mol
 - c. 22.24 mol
 - d. 44.48 mol
10. The number of moles of H_2 that can be produced from 100.0 g of H_2O is:
- a. 5.56 mol
 - b. 11.12 mol
 - c. 22.24 mol
 - d. 44.48 mol



Section Quiz Introduction To Stoichiometry Answers

Ying-Ying Zheng



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