

AP Chemistry Chapter 1-4 Test Review

1. What are the symbols for the elements: lead, silver, gold, potassium, mercury, iron, and sodium.
2. What are the SI units for: mass, length, volume, energy, time?
3. The density of aluminum is 2.7 g/cm^3 . A piece of aluminum that occupies a volume of 21.4 cm^3 would have a mass of?
4. The number 1.00430 has _____ significant figures.
A) 2 B) 3 C) 5 D) 6 E) 4
5. Identify the following as a chemical reaction or physical change:
A) Composting of grass
B) Evaporation of water
C) Rusting of a nail
D) Combustion of methane
E) Cutting paper
6. Identify which of the following are intensive physical properties and which are extensive:
A) Volume B) Mass C) Density D) Melting Point E) Color
7. What does the atomic number and atomic mass tell about the atom?
8. How do isotopes differ? Write the shorthand notation for two isotopes of an element.
9. Which elements are metals, which are metalloids and which are nonmetals?
10. What happens when a metal reacts with a nonmetal? What about a nonmetal and non metal?
11. What is the empirical formula for $\text{C}_4\text{H}_8\text{O}_2$?
12. What are the typical oxidation numbers for each family?
13. Write the formula for Aluminum reacting with Oxygen.
14. Name the following substances: HI , CO_2 , MgCO_3 , H_3PO_4 , SF_6 , MnO
15. What are the major contributions of the following scientists to the atomic theory: Dalton, Thompson, Rutherford, Goldstein, Millikan, and Chadwick
16. How many electrons, neutrons and protons does ^{14}C have?
17. What is the smallest particle of an atom?

Ap Chemistry Chapter 1 Test Wikispaces

Mary Millhollon, Richard H. Langley



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AP Chemistry For Dummies Peter J. Mikulecky, Michelle Rose Gilman, Kate Brutlag, 2008-11-13 A practical and hands on guide for learning the practical science of AP chemistry and preparing for the AP chem exam Gearing up for the AP Chemistry exam AP Chemistry For Dummies is packed with all the resources and help you need to do your very best Focused on the chemistry concepts and problems the College Board wants you to know this AP Chemistry study guide gives you winning test taking tips multiple choice strategies and topic guidelines as well as great advice on optimizing your study time and hitting the top of your game on test day This user friendly guide helps you prepare without perspiration by developing a pre test plan organizing your study time and getting the most out of your AP course You ll get help understanding atomic structure and bonding grasping atomic geometry understanding how colliding particles produce states and so much more To provide students with hands on experience AP chemistry courses include extensive labwork as part of the standard curriculum This is why the book dedicates a chapter to providing a brief review of common laboratory equipment and techniques and another to a complete survey of recommended AP chemistry experiments Two full length practice exams help you build your confidence get comfortable with test formats identify your strengths and weaknesses and focus your studies You ll discover how to Create and follow a pretest plan Understand everything you must know about the exam Develop a multiple choice strategy Figure out displacement combustion and acid base reactions Get familiar with stoichiometry Describe patterns and predict properties Get a handle on organic chemistry nomenclature Know your way around laboratory concepts tasks equipment and safety Analyze laboratory data Use practice exams to maximize your score Additionally you ll have a chance to brush up on the math skills that will help you on the exam learn the critical types of chemistry problems and become familiar with the annoying exceptions to chemistry rules Get your own copy of AP Chemistry For Dummies to build your confidence and test taking know how so you can ace that exam

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