

Chapter 6 Practice Test: Chemical Bonding

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. The charge on an ion is
a. always positive.
b. always negative.
c. either positive or negative.
d. zero.
- _____ 2. According to the octet rule, a calcium atom has a tendency to
a. lose one electron.
b. lose two electrons.
c. gain one electron.
d. gain two electrons.
- _____ 3. If a compound forms by ionic bonding, which is *not* true?
a. A positively charged atom or group of atoms attracts a negatively charged atom or group of atoms.
b. The net charge of the compound is zero.
c. The compound contains just two atoms, each of opposite charge.
d. Several ions group together in a tightly packed structure.
- _____ 4. The only property listed that is *not* characteristic of ionic compounds is
a. high melting point.
b. hardness.
c. lack of crystal structure.
d. good conductor of electricity.
- _____ 5. Which formula listed below represents a polyatomic ion?
a. NH_4^+
b. H_2SO_4
c. NaCl
d. H_2O
- _____ 6. The melting points of ionic compounds are higher than the melting points of molecular compounds because
a. ionic substances tend to vaporize at room temperature.
b. ionic substances are brittle.
c. attractive forces between ions are greater than the attractive forces between molecules.
d. the numbers of positive and negative charges are equal in an ionic compound.

Chapter 6 Chemical Bonding Test

Christian G. Meyer



Chapter 6 Chemical Bonding Test:

Embracing the Melody of Expression: An Psychological Symphony within **Chapter 6 Chemical Bonding Test**

In a global eaten by screens and the ceaseless chatter of immediate interaction, the melodic splendor and emotional symphony produced by the written term frequently diminish in to the background, eclipsed by the constant noise and disruptions that permeate our lives. However, located within the pages of **Chapter 6 Chemical Bonding Test** an enchanting fictional treasure overflowing with natural thoughts, lies an immersive symphony waiting to be embraced. Constructed by an elegant musician of language, this charming masterpiece conducts visitors on a mental journey, skillfully unraveling the concealed melodies and profound affect resonating within each carefully crafted phrase. Within the depths of this moving analysis, we will investigate the book is key harmonies, analyze their enthralling writing style, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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