

# Compounds, Ions, and Molecules

## Chapter Preview

If elements are the “building blocks of matter,” then what kinds of things do they build? The answer is found in the world within and around you. Elements combine to make compounds. For example, copper and sulfur combine to form a compound called copper(II) sulfide. This compound is found in copper ore. Mining companies dig up the ore, and after many stages of processing, end up with the element copper.

In this chapter, we will review some fundamental ideas about compounds and introduce some new ideas. You will learn how different types of compounds form, and how they are described in the language of chemistry.

### KEY IDEAS

Compounds result when elements bond together in fixed proportions.

Bohr diagrams can illustrate how ions form.

Bonding can involve electron transfer (ionic) or electron sharing (covalent).

Rules for writing chemical formulas and for naming ionic compounds are based on ion charge balances.

Rules for writing chemical formulas and for naming molecular compounds are based on a prefix system.

## TRY THIS: Elements and Compounds Around You

**Skills Focus:** interpreting data

Pure substances are everywhere in your daily life. In this activity, you will consider a list of a few pure substances—elements and compounds—commonly found in your kitchen. Remember that pure substances contain only one kind of particle.

**Table 1** Some Pure Substances in the Kitchen

oxygen (in air)	aluminum (aluminum foil)
nitrogen (in air)	sodium chloride (table salt)
carbon dioxide (in air)	sucrose (table sugar)
ammonia (in spray cleaner)	water (from the tap)
tungsten (in light bulbs)	methane (natural gas for a gas stove)
sodium bicarbonate (baking soda)	polyethylene (plastic bags)

1. Make one list of the elements in Table 1 and another list of the compounds.
  - A. How did you determine which of the pure substances were elements? How did you determine which of the pure substances were compounds?
  - B. Compare your lists with other students in the class.
  - C. If you examined all of the pure substances in a kitchen, which list do you think would be longer? Why do you think that list would be longer?
  - D. Suggest one other example of an element and one of a compound commonly found in a kitchen.

# Chapter 7 Compounds Ions And Molecules

**Marcel A. Müller**



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## **Table of Contents Chapter 7 Compounds Ions And Molecules**

1. Understanding the eBook Chapter 7 Compounds Ions And Molecules
  - The Rise of Digital Reading Chapter 7 Compounds Ions And Molecules
  - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 7 Compounds Ions And Molecules
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chapter 7 Compounds Ions And Molecules
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 7 Compounds Ions And Molecules
  - Personalized Recommendations
  - Chapter 7 Compounds Ions And Molecules User Reviews and Ratings
  - Chapter 7 Compounds Ions And Molecules and Bestseller Lists
5. Accessing Chapter 7 Compounds Ions And Molecules Free and Paid eBooks
  - Chapter 7 Compounds Ions And Molecules Public Domain eBooks
  - Chapter 7 Compounds Ions And Molecules eBook Subscription Services
  - Chapter 7 Compounds Ions And Molecules Budget-Friendly Options
6. Navigating Chapter 7 Compounds Ions And Molecules eBook Formats
  - ePub, PDF, MOBI, and More
  - Chapter 7 Compounds Ions And Molecules Compatibility with Devices
  - Chapter 7 Compounds Ions And Molecules Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chapter 7 Compounds Ions And Molecules
  - Highlighting and Note-Taking Chapter 7 Compounds Ions And Molecules
  - Interactive Elements Chapter 7 Compounds Ions And Molecules
8. Staying Engaged with Chapter 7 Compounds Ions And Molecules

- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chapter 7 Compounds Ions And Molecules
9. Balancing eBooks and Physical Books Chapter 7 Compounds Ions And Molecules
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chapter 7 Compounds Ions And Molecules
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Chapter 7 Compounds Ions And Molecules
- Setting Reading Goals Chapter 7 Compounds Ions And Molecules
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chapter 7 Compounds Ions And Molecules
- Fact-Checking eBook Content of Chapter 7 Compounds Ions And Molecules
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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