

**STRENGTH 200** – **Introduction to Chemistry – Laboratory Experiments – Lighting, 2000**

- Experiment:** When there is a reaction between a substance and oxygen, the reaction is called a combustion reaction. This is the case for the reaction between hydrogen and oxygen to form water.
- Objective:** The purpose of this experiment is to demonstrate the reaction between hydrogen and oxygen to form water. The reaction is exothermic and produces a flame.
- Background:** Hydrogen is a colorless, odorless gas that is highly flammable. It is the lightest element and is produced by the electrolysis of water.
- Key Concepts:** The reaction between hydrogen and oxygen to form water is a combustion reaction. The reaction is exothermic and produces a flame. The reaction is also a redox reaction.
- Procedure:** The purpose of this experiment is to demonstrate the reaction between hydrogen and oxygen to form water. The reaction is exothermic and produces a flame. The reaction is also a redox reaction.
- Results:** The reaction between hydrogen and oxygen to form water is a combustion reaction. The reaction is exothermic and produces a flame. The reaction is also a redox reaction.
- Conclusion:** The reaction between hydrogen and oxygen to form water is a combustion reaction. The reaction is exothermic and produces a flame. The reaction is also a redox reaction.
- References:** The following references were used in the preparation of this experiment: [Introduction to Chemistry, 2000](#), [Chemistry, 2000](#), and [Laboratory Experiments, 2000](#).

# Chem 110 Introductory Chemistry Lecture Syllabus Spring

**Clemens Wendtner**



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