

# STOICHIOMETRY WORKSHEET

1. Calculate the number of moles of NaOH that are needed to react with 500 g of H<sub>2</sub>SO<sub>4</sub> according to the following equation:



2. Calculate the mass of NH<sub>3</sub> that can be produced from the reaction of 125 g of NCl<sub>3</sub> according to the following equation:



3. Identify the limiting reactant and determine the mass of CO<sub>2</sub> that can be produced from the reaction of 25 g of C<sub>3</sub>H<sub>8</sub> with 75 g of O<sub>2</sub> according to the following equation:



4. How many grams of SO<sub>2</sub> are produced when 152 g of CS<sub>2</sub> reacts with 48.0 g of O<sub>2</sub> according to the following equation:



# Chemistry Chapter 12 Stoichiometry Worksheet Answers

**Kathleen Armour**



**Chemistry Chapter 12 Stoichiometry Worksheet Answers:**

## Unveiling the Magic of Words: A Report on "**Chemistry Chapter 12 Stoichiometry Worksheet Answers**"

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