

## Chemical Analysis – What can you Remember?

1. What colour precipitate do chloride ions produce?
  - a) White
  - b) Cream
  - c) Yellow
  - d) Green
2. What colour do sodium ions go in a flame test?
  - a) Red
  - b) Yellow
  - c) Brick-red
  - d) Lilac
3. Which two solutions are added to test for the presence of halide ions?
  - a) Silver chloride and nitric acid
  - b) Barium chloride and hydrochloric acid
  - c) Silver nitrate and nitric acid
  - d) Barium chloride and sulphuric acid
4. What is the test for oxygen gas?
  - a) Lit splint goes out with a squeaky pop
  - b) Damp blue litmus paper goes red
  - c) Damp red litmus paper goes blue
  - d) Glowing splint relights
5. What colour precipitate do iodide ions form?
  - a) White
  - b) Cream
  - c) Yellow
  - d) Green
6. Which ion produces a brick-red colour in a flame test?
  - a) Sodium
  - b) Calcium
  - c) Copper
  - d) Lithium
7. What is the test for carbon dioxide gas?
  - a) Limewater turns cloudy
  - b) Damp blue litmus turns red
  - c) A glowing splint relights
  - d) A lit splint goes out with a squeaky pop
8. When sodium hydroxide is added to a solution containing iron (III) ions, what colour precipitate forms?
  - a) White
  - b) Blue
  - c) Green
  - d) Red/brown
9. What is the test for chlorine gas?
  - a) Damp blue litmus paper turns red and is then bleached white
  - b) Red litmus  turns blue
  - c) A glowing splint relights
  - d) Limewater will turn cloudy
10. Copper (II) ions form what colour precipitate test?
  - a) Blue
  - b) Green
  - c) Red/brown
  - d) Cream
11. Which two solutions are added to test for the presence of sulphate ions?
  - a) Silver chloride and nitric acid
  - b) Barium chloride and sulphuric acid
  - c) Silver nitrate and nitric acid
  - d) Barium chloride and hydrochloric acid
12. What colour do potassium ions go in a flame test?
  - a) Lilac
  - b) Red
  - c) Yellow
  - d) Brick-red
13. What is the test for hydrogen gas?
  - a) Lit splint goes out with a squeaky pop
  - b) Glowing splint relights
  - c) Damp blue litmus paper turns red and then is bleached white
  - d) Limewater goes cloudy
14. Which ion produces a red colour in a flame test?
  - a) Sodium
  - b) Calcium
  - c) Copper
  - d) Lithium
15. What colour precipitate do bromide ions form?
  - a) White
  - b) Cream
  - c) Yellow
  - d) Green
16. What needs to be added to test for the presence of carbonate ions?
  - a) Silver nitrate
  - b) Universal indicator
  - c) Litmus paper
  - d) Dilute acid
17. If sodium hydroxide is added to a compound containing ammonium ions, what gas is produced?
  - a) Oxygen
  - b) Ammonia
  - c) Carbon dioxide
  - d) Hydrogen
18. What colour precipitate does barium sulphate form?
  - a) White
  - b) Green
  - c) Yellow
  - d) Cream
19. What is needed to carry out a flame test?
  - a) Splint
  - b) Glass rod
  - c) Spatula
  - d) Nichrome/platinum wire

Your score out of 19 =  
Percentage =

# Chemistry Unit 2 Test Answers

**A Gutmann**



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