

Nuclear Fission and Fusion

Why?

Fission and fusion are two processes that give rise to energy of an atom. Nuclear fission provides the energy to produce power plants and fusion will be used in the near future. The use of fusion in power plants can help produce clean energy. However, the energy produced by fission of isotopes of uranium, the fuel, is used to produce energy that is distributed to other uses and power plants are the source of

Learning Objectives

1. Calculate and compare nuclear fission and fusion reactions.
2. Understand the conservation laws that apply to nuclear reactions.

Success Criteria

1. Show working demonstrating the correct solution.
2. Identify correct conservation equations of fission or fusion.

Vocabulary

- Fusion
- Fission
- Conservation
- Reaction

Prerequisites

1. Knowledge of mass and energy.
2. Knowledge of wave number and atomic number.
3. Atomic symbols including mass number, proton number, and charge.

Introduction

Transmutation is the transformation of the nucleus of an atom as they the mass is changed from the nucleus into a different element. This can be accomplished through many types of reactions, including fusion and fission.

Fission, particularly of uranium, is the only one that produces energy. Fission is used in power plants and (2) nuclear charge is conserved.

Energy and mass are conserved, but not in the conventional understanding. Einstein's equation, $E=mc^2$.

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TD Snyder



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